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# Modeling of the CO<sub>2</sub> Plume

Ramon Carbonell





CIUDEN: Spanish government foundation to promote, among other things, Carbon Capture and Storage



MUSTANG: EU 7th FP project: “A MUltiple Space and Time scale Approach for the quaNtification of deep saline formations for CO2 storaGe” (2009-2013)



Hontomin is the Tech Demonstration Plant of the Compostilla OXYCFB300 EEPR project, run by ENDESA, in collaboration with CIUDEN and FOSTER-WHEELER



EEPR “European Energy Programme for Recovery” facilitates investments on infrastructure and technology projects in the energy sector; helps improve the security of supply of the Member States and, promotes implementation of the 20/20/20 objectives for 2020.



CSIC is the “Spanish Agency for Scientific Research”, a network of research institutes.

# The Players

## The Research Team:

Jesús Carrera<sup>1</sup>, Orlando Silva<sup>12</sup>, Victor Vilarrasa<sup>3</sup>, Marco Dentz<sup>2</sup>, Bruno Cubillo<sup>23</sup>, Francesca de Gaspari<sup>23</sup>, Lurdes Martínez<sup>3</sup>, Albert Nardi<sup>3</sup>, Maarten Saaltink<sup>3</sup>, Luit Jan Slooten<sup>3</sup>, Josep Soler<sup>2</sup>, Carlos Ayora<sup>2</sup>, Jordi Cama<sup>2</sup>, Gabriela Dávila<sup>3</sup>, María García<sup>3</sup>, Tobias Rötting<sup>23</sup>

## The Journalist: R. Carbonell<sup>1,4</sup>



1. Subprogram of CO2 Storage, Energy City Foundation



2. Institute of Environmental Assessment and Water Research, CSIC

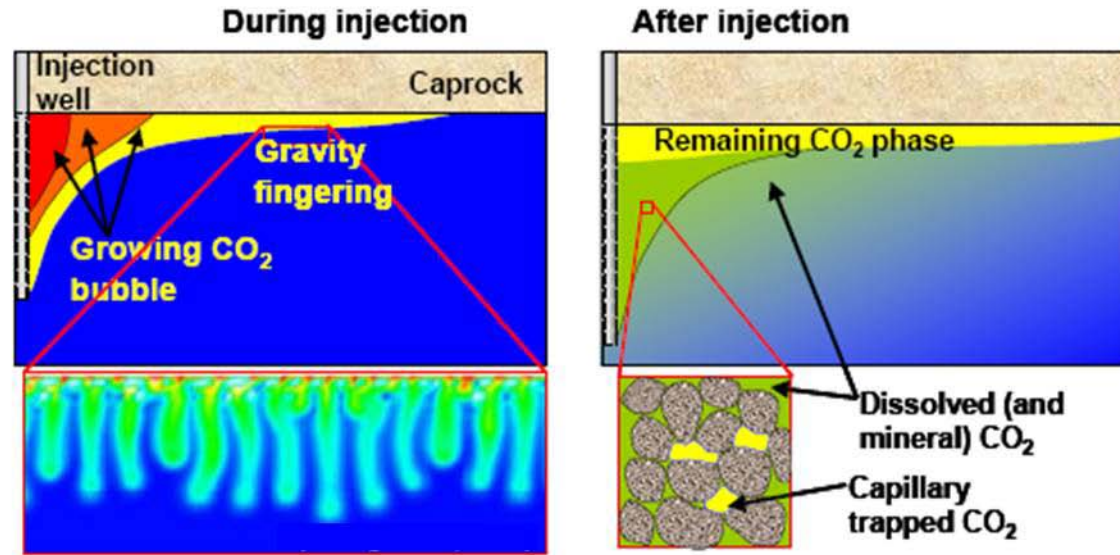


3. Hydrogeology Group, UPC



4. Institute for Earth Sciences Jaume Amera, Spanish Agency for Scientific Research, CSIC

# Main issues



- ❑ Movement of the CO<sub>2</sub>-rich phase within the geological formations (long-term sequestration).
- ❑ Understanding of different mechanisms of trapping: structural, capillary, solute and mineral.
- ❑ CO<sub>2</sub> dissolution into the aqueous phase (brine) (solute trapping: ideally as bicarbonate, but also carbonic acid).
- ❑ Interaction between these solutes and minerals, and the ability of the geological formation to neutralize the solution and mineralize the CO<sub>2</sub>.
- ❑ Coupled hydro-mechanical effects (porosity and permeability changes) caused by the evolution of fluid pressures during and after CO<sub>2</sub> injection.
- ❑ Management of CO<sub>2</sub> at surface and injection strategies.

# Objectives

- ❑ Development of numerical, analytical and semi-analytical tools.
- ❑ Design and planning of characterization tests (hydro-thermo-chemical-mechanical) and CO<sub>2</sub> injection tests that will be undertaken at the Technological Demonstration Plant (TDP) of Hontomín.
- ❑ Interpretation of characterization tests:
  - Knowledge of chemical and structural composition, heterogeneity of the geological formations.
  - To know the porosity changes.
  - To determine the hydraulic connectivity between reservoir and caprock.
  - To know the dynamics of CO<sub>2</sub> migration within the reservoir.
  - Hydro-mechanical response of the system: evolution of pressures in the reservoir as CO<sub>2</sub> is injected; mechanical resistance of caprock; risks of leakages.
  - To know if the CO<sub>2</sub> will be effectively trapped.
- ❑ Modeling also can contribute to the development of CCS integration models (capture-transport-storage), and new technologies (e.g., injection strategies to improve the efficiency in the short term and the effectiveness in the long term).

# Methodology (tools)

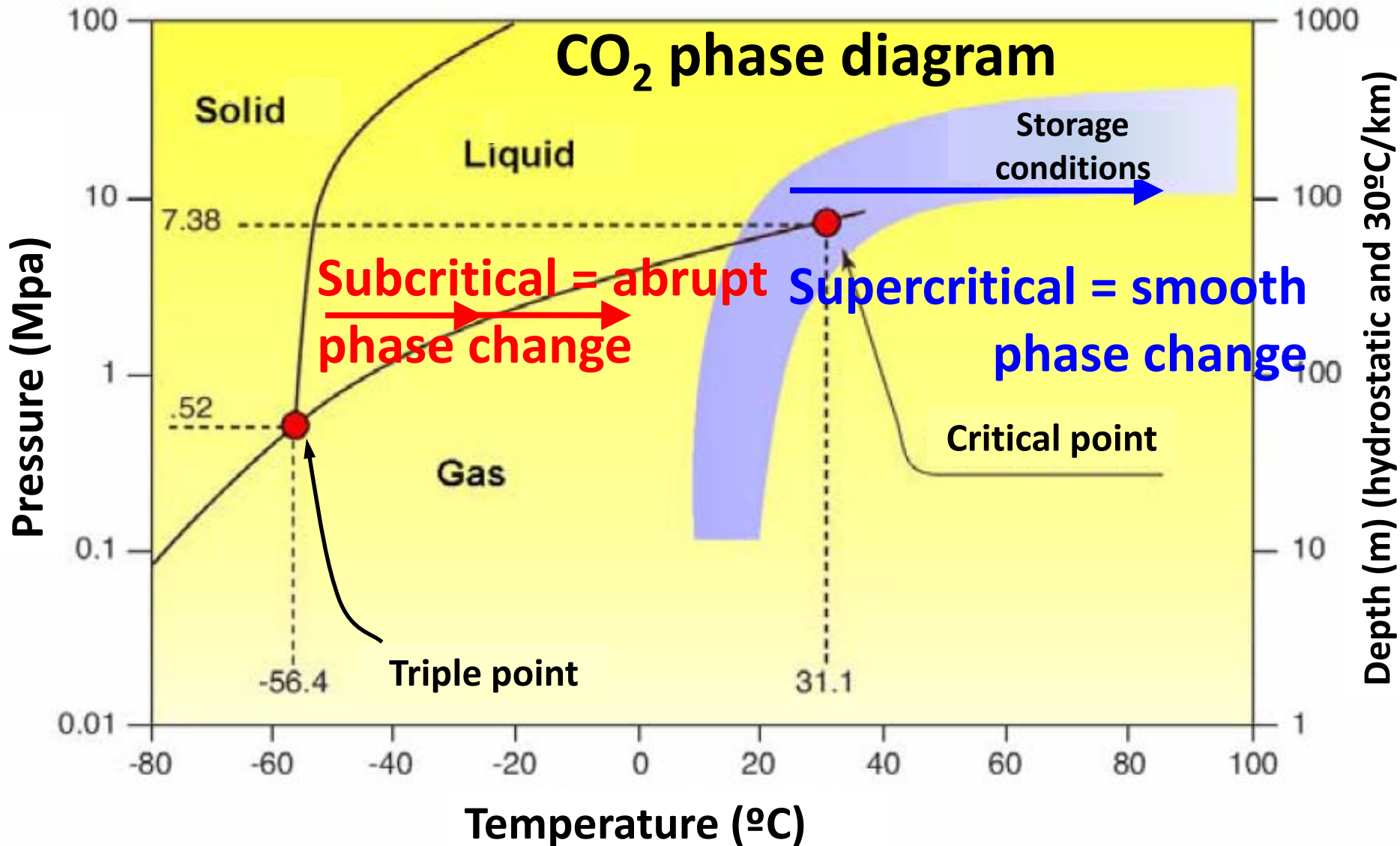
## Modeling of multi-phase flow and multi-component reactive transport

- ❑ Thermodynamics of CO<sub>2</sub> (gas, liquid and supercritical states) and brine.
- ❑ Thermodynamics of multi-component gaseous mixtures (EoS, fugacity coefficients, equilibrium constants).
- ❑ Analytical solutions and non-local formulations (both in time and space), upscaling: MRMT, R-MRMT.
- ❑ Use of existing codes
  - INVERT
  - TRANSIN
  - CODE\_BRIGHT
  - RETRASO
  - COGERE
  - KRINET
- ❑ Code development: object oriented programming (PROOST, CHEPROO): FORTRAN, C++, MatLab.

# Key issues

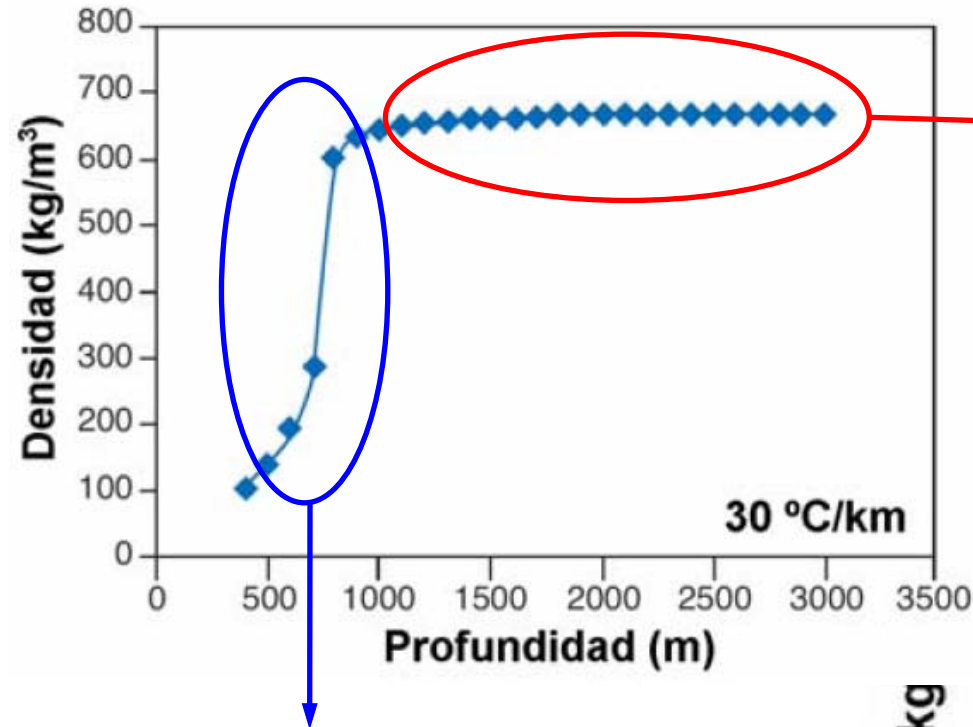
- Fluid mechanics (hydraulic) issues
  - Basic description
  - Scoping calculations for: Pressure impact (magnitude and extent) and CO<sub>2</sub> bubble
  - Hydraulic characterization. Preferential flow paths
- Geo Chemical issues (coupled to hydraulic)
  - CO<sub>2</sub> Dissolution and fingering
  - Mineral dissolution (and precipitation?)
- Mechanical issues
  - Stability
  - Microseismic activity
- Hontomin

# CO<sub>2</sub> supercritical?



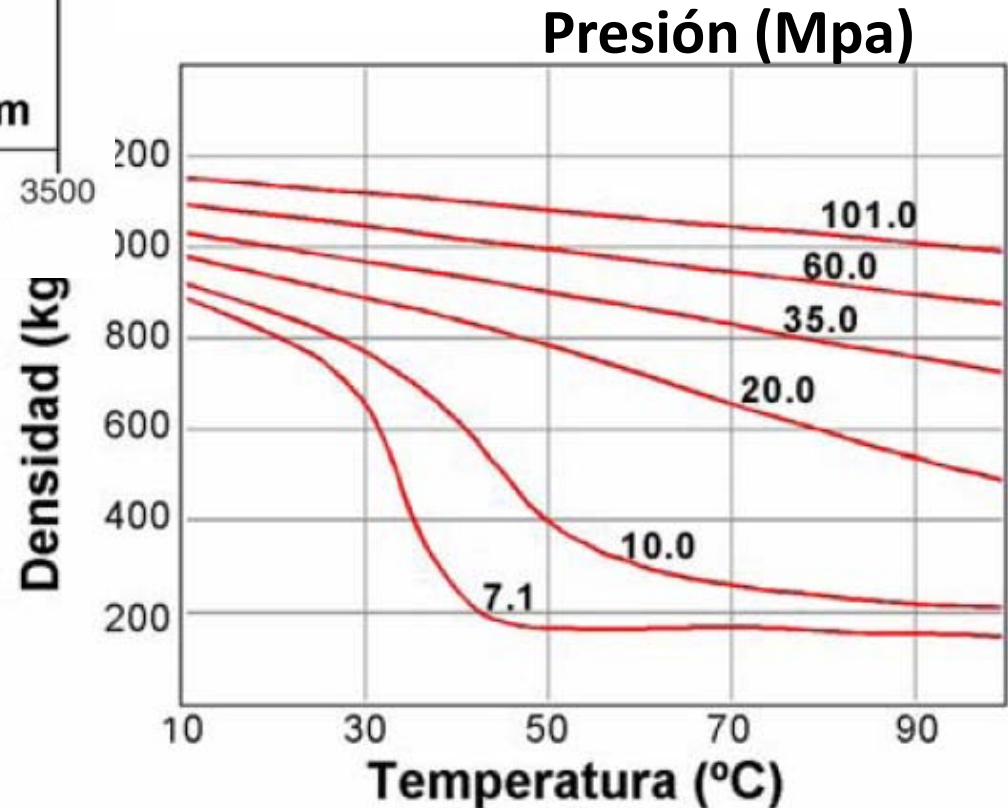


# CO2 supercritical?

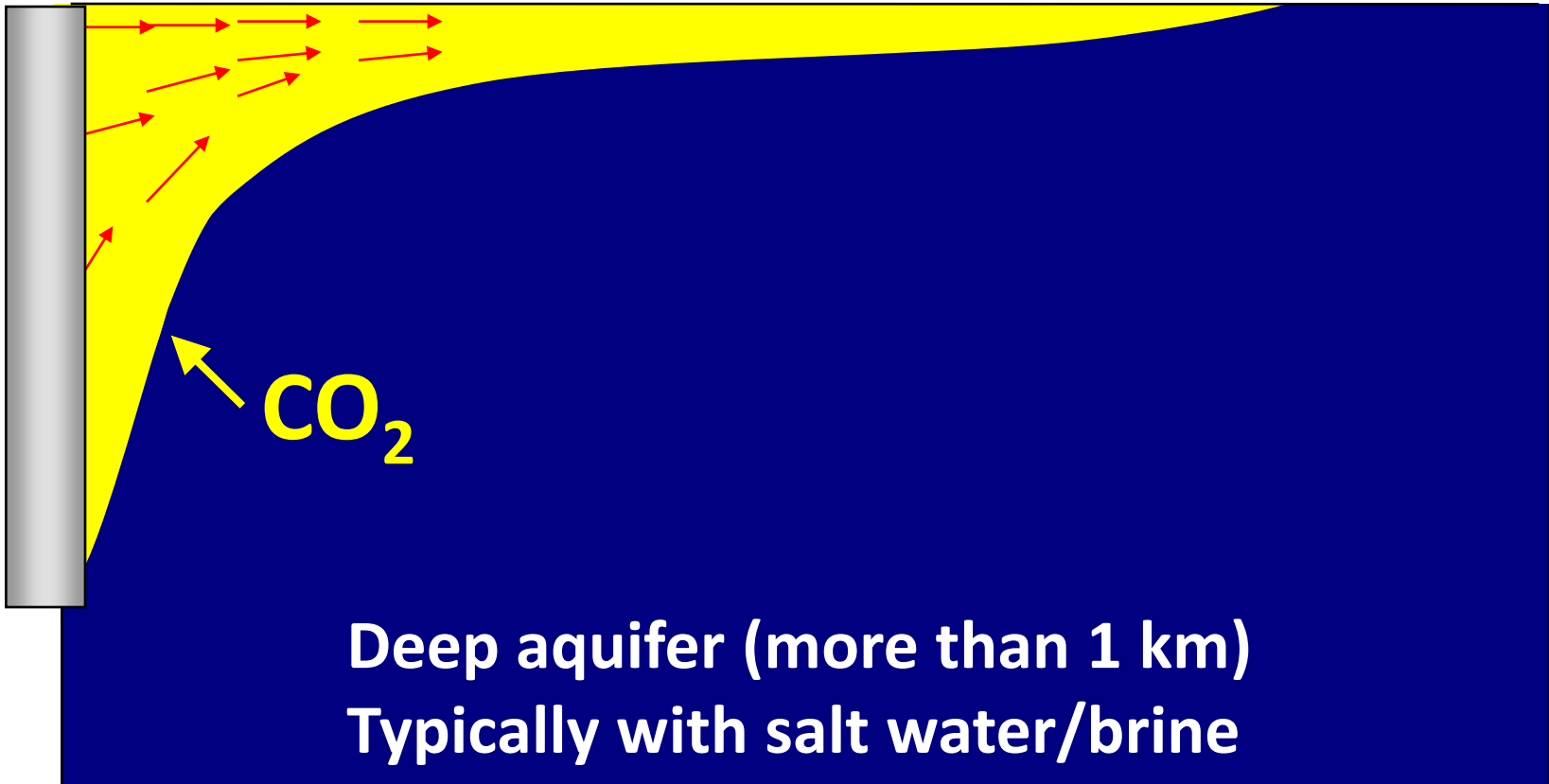


For high P, high density low compressibility (like a liquid)

For low P & T, low density & high compressibility (like a gas)

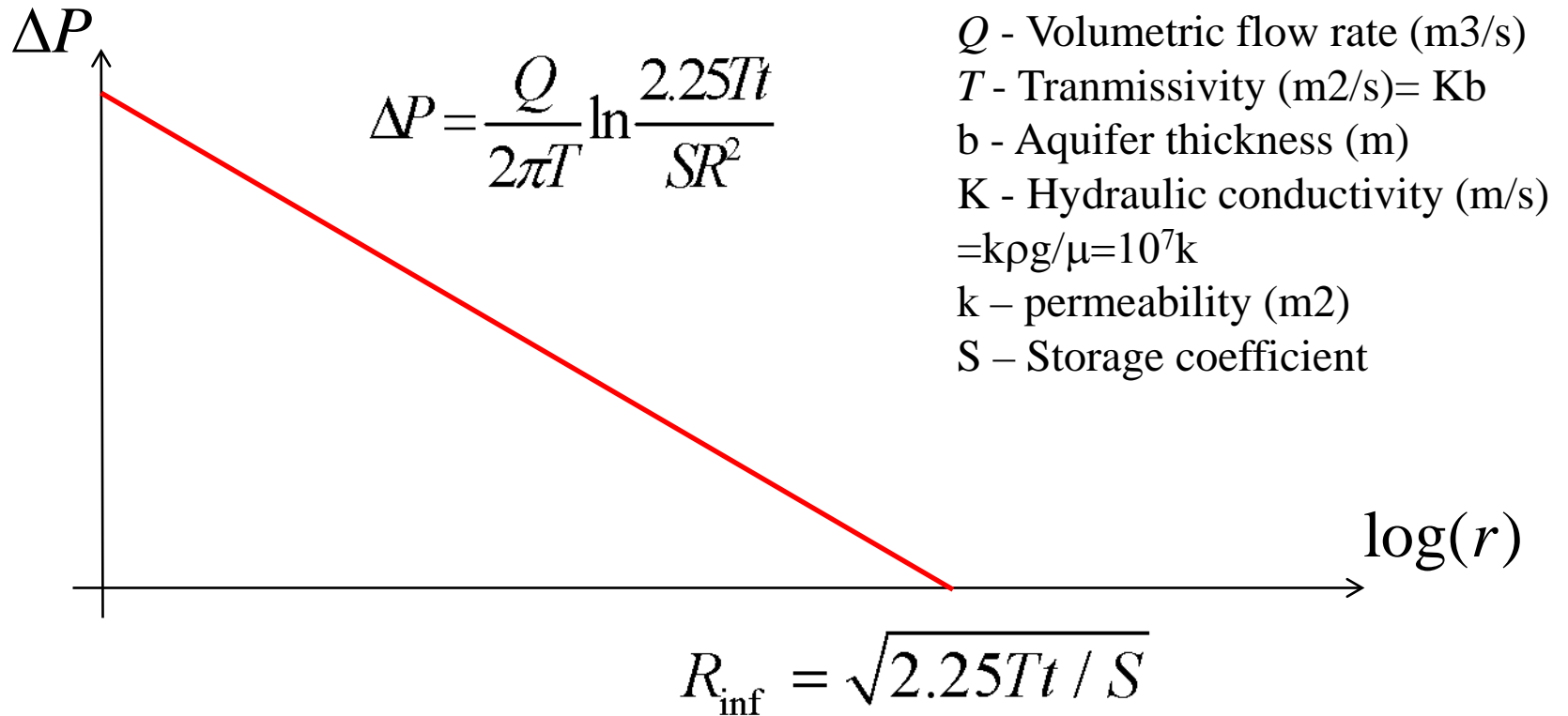


# Buoyancy and flow



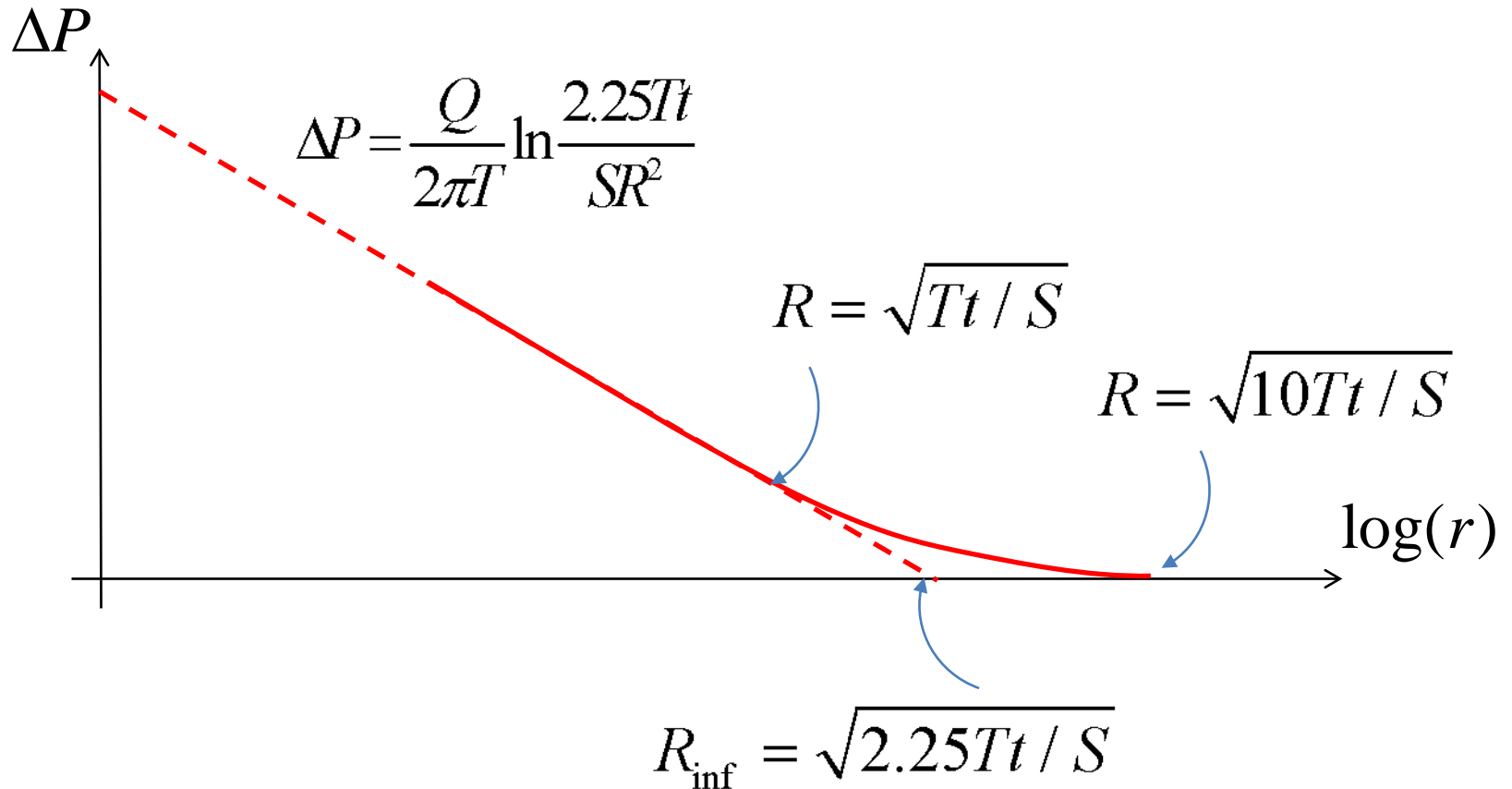
# Scoping calculations: Regional pressure buildup

Most of the pressure build-up is associated to the resistance of water to be displaced



# Scoping calculations: Regional pressure buildup

Most of the pressure build-up is associated to the resistance of water to be displaced



# Scoping calculations (regional): the bad news

## Pressure buildup and Radius of influence

Assume a 1Mt/year injection rate

If compressible and permeable formation ( $k=10^{-12} \text{ m}^2$ ,  $S_s=10^{-5} \text{ m}^{-1}$ )

t (days)	10 days	30 years	
Rinf (km)	1.39	46.1	Large region affected
$\Delta P$ (bar) (r=100m)	1.69	3.94	Risk brine migration

If low compressibility and permeability formation ( $k=10^{-14} \text{ m}^2$ ,  $S_s=10^{-6} \text{ m}^{-1}$ )

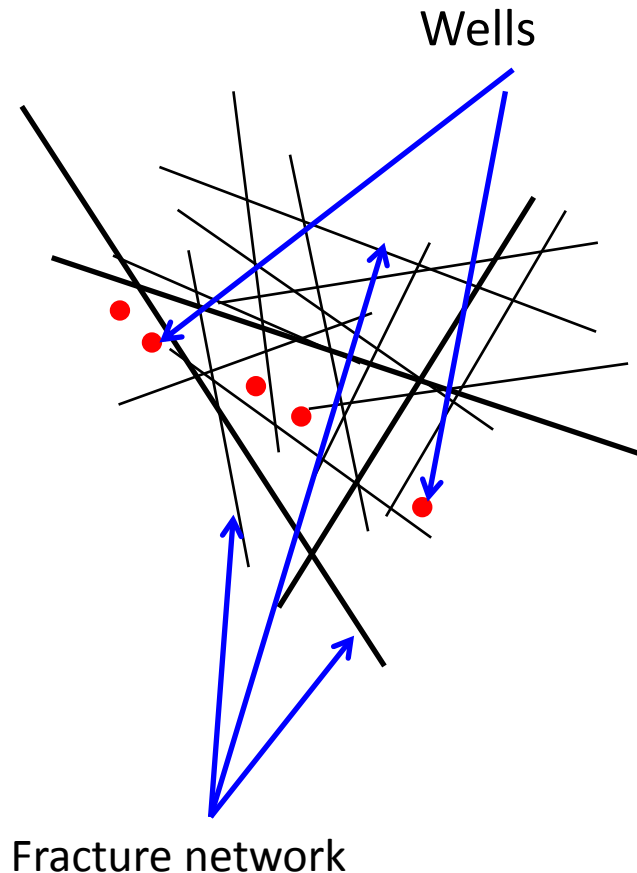
t (days)	10 days	30 years	
Rinf (km)	0.44	14.6	
$\Delta P$ (bar) (r=100m)	95.2	320	Unbearable pressure

Observations:

- 1) Characterization (geologic and hydraulic) needed
- 2) Very likely, brine pressure control will be needed

But, do not go away yet!

# Cause (and solution) of scale effects



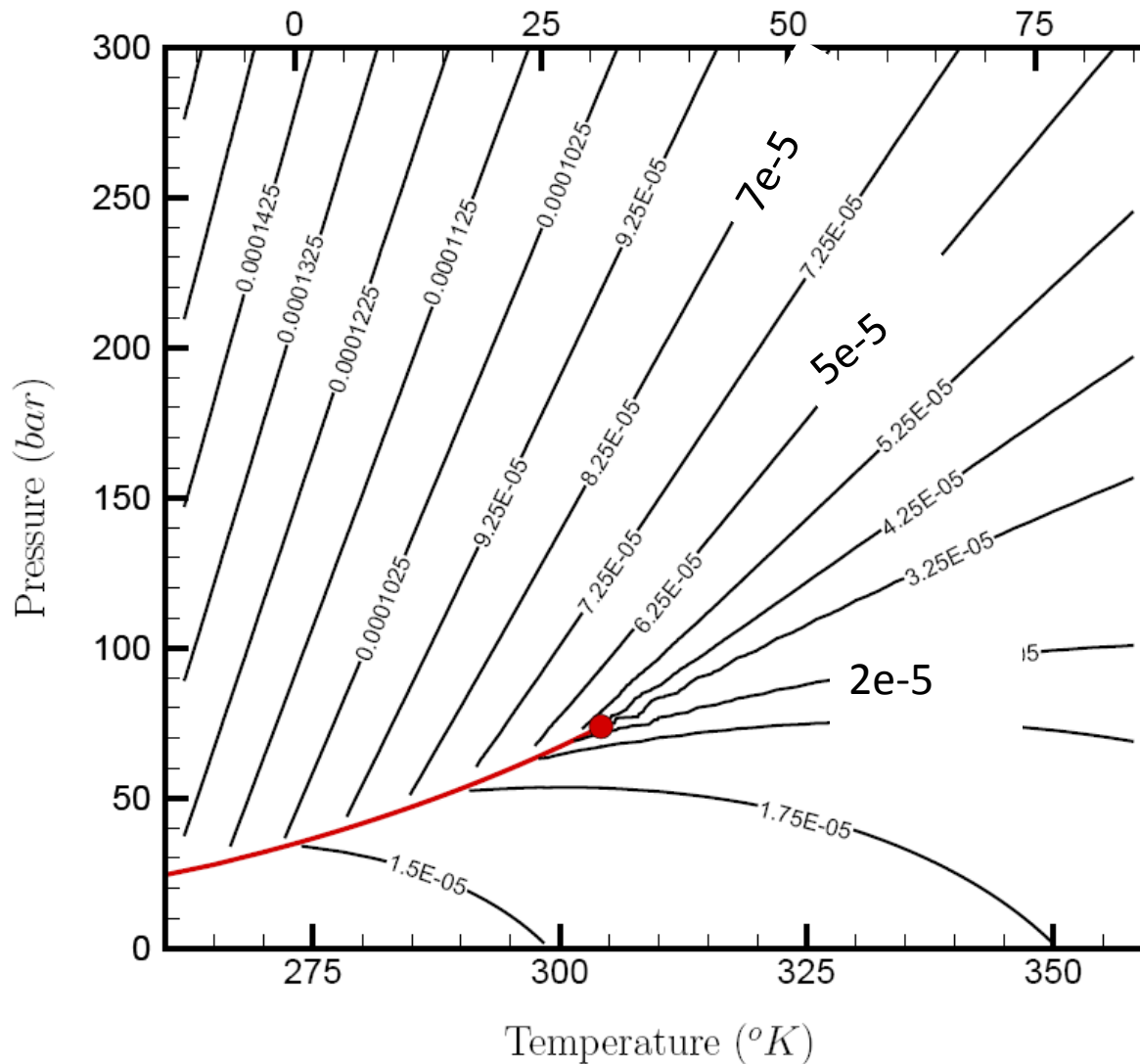
Large scale permeability, which is derived from hydraulic tests, is controlled by large fractures

The effective permeability of every well depends on whether or not a conducting fracture is intersected. Chances of hitting a subvertical fracture are low.

Wells act as if they were poorly built

**Solution: Long horizontal wells**

# Low viscosity

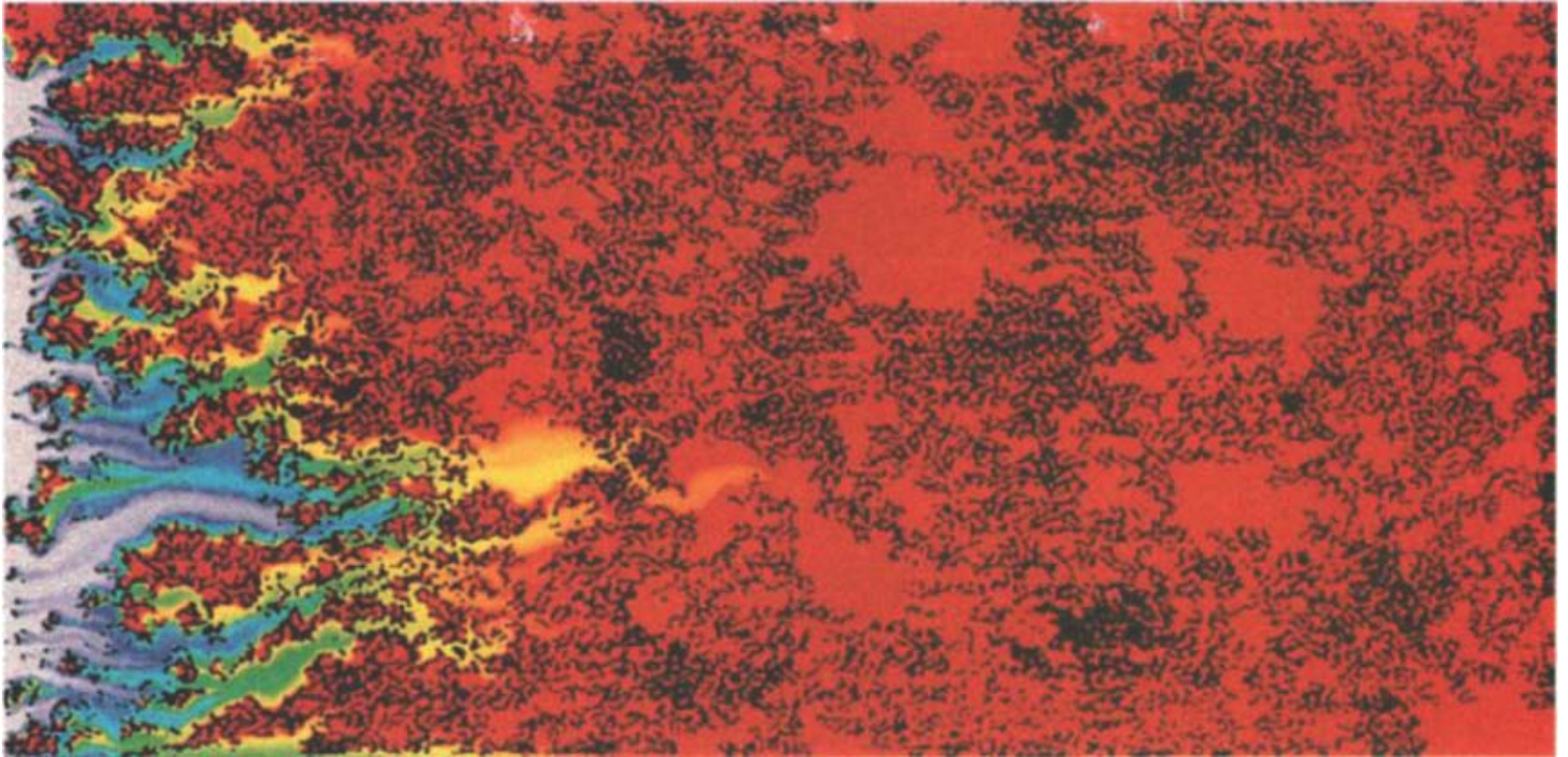


Pure CO<sub>2</sub> viscosity (Pa·s)

CO<sub>2</sub> viscosity is some 10-20 times smaller than water

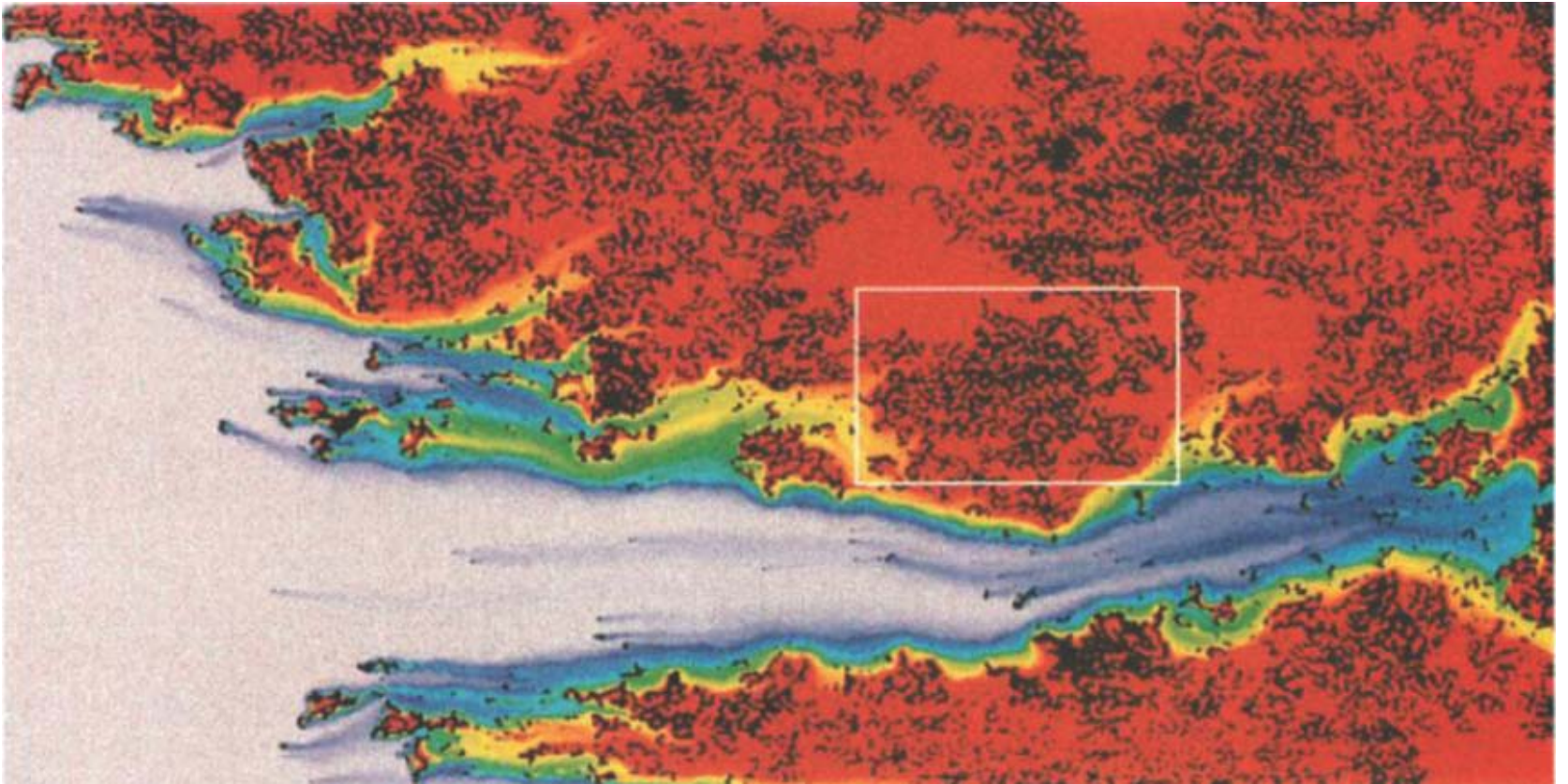
Water viscosity 0.001 Pa·s

# Viscous fingering

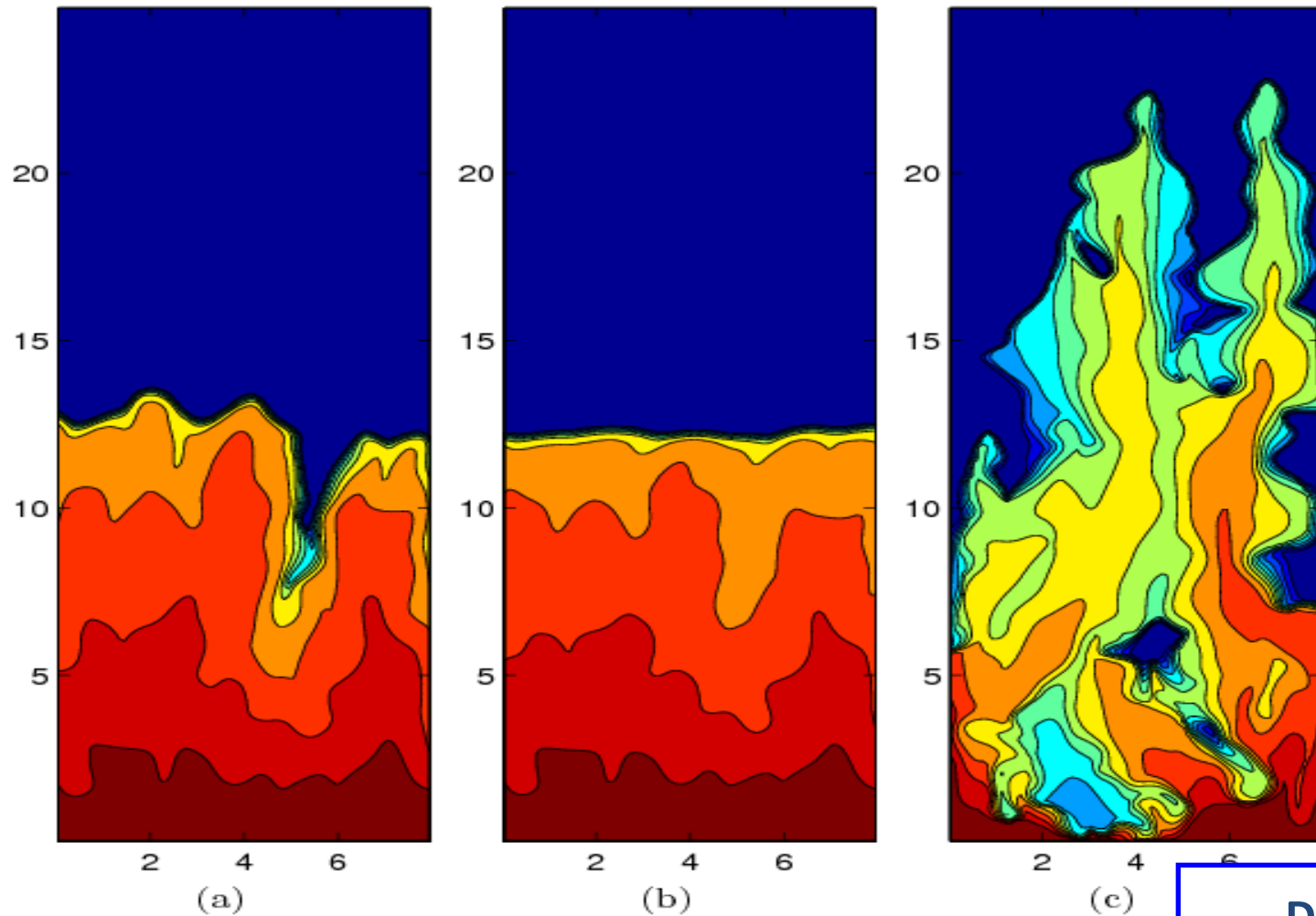




# Viscous fingering



# Fingering increases by superposition of buoyancy and viscosity effects

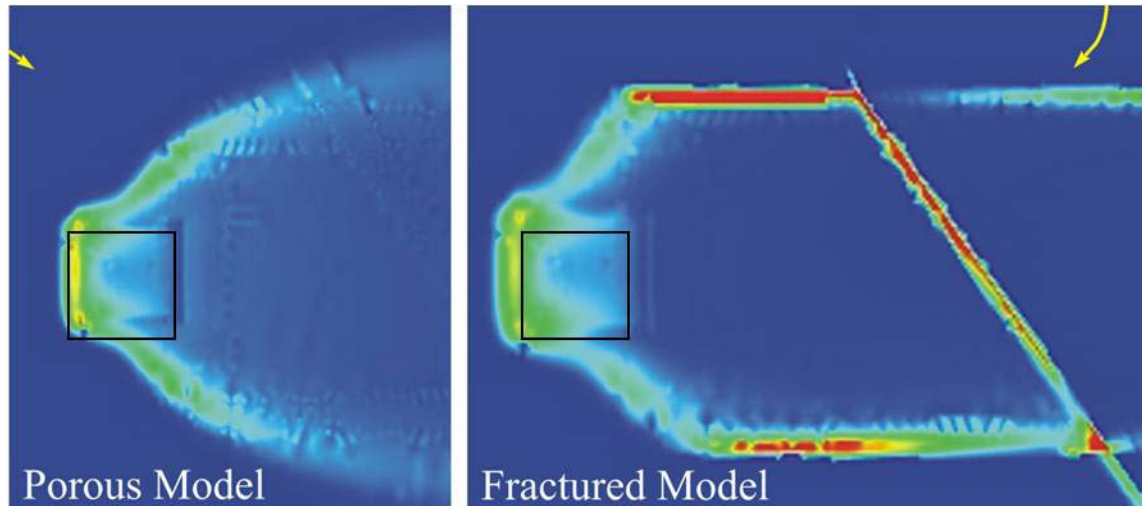


Diogo

# R-MRMT and MRMT-R

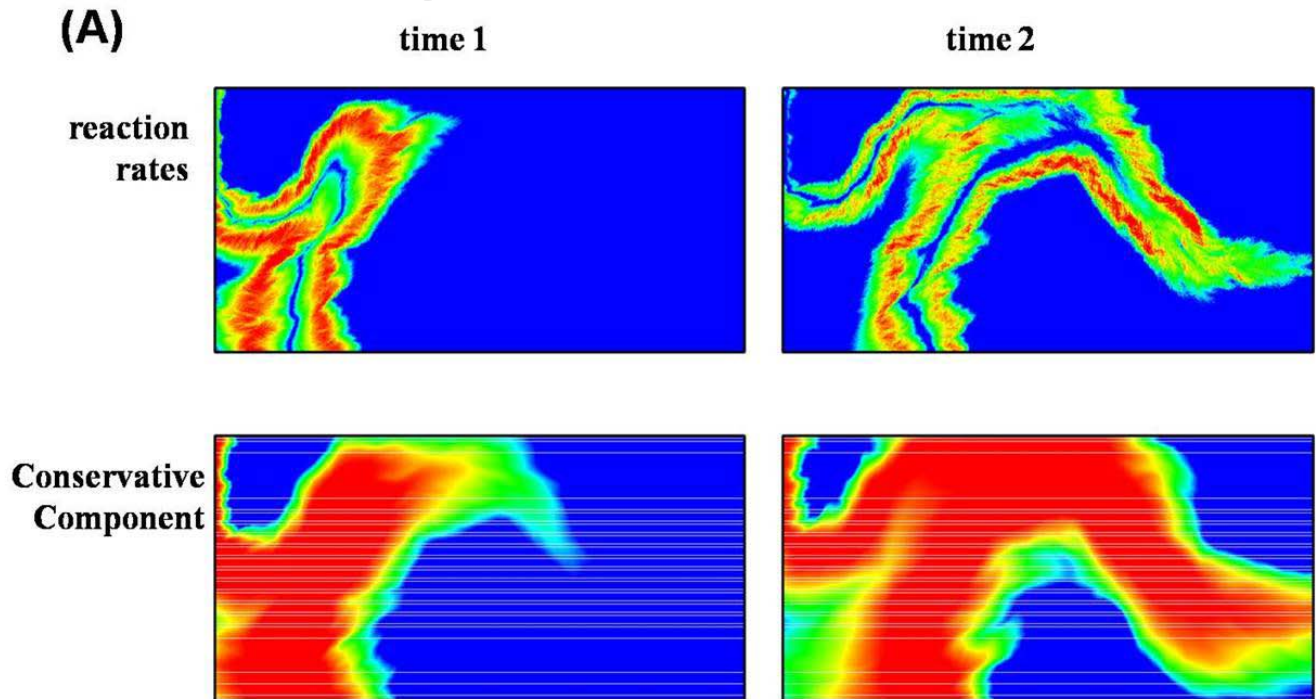
1. Mixing induces reaction

2. The presence of mobile and immobile zones induces mixing

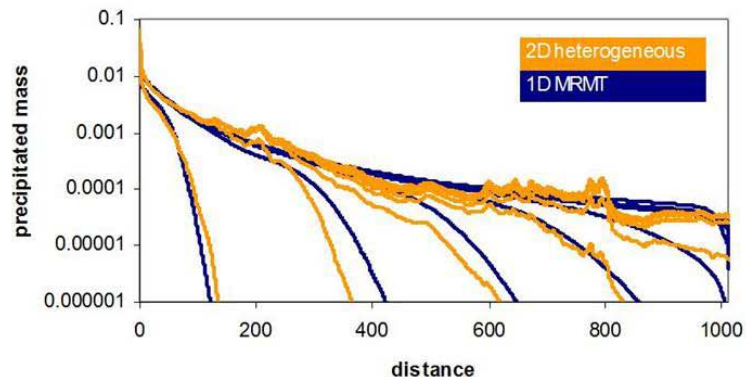


3. To take into account reaction induced by mixing: R-MRMT (Reactive Multi Rate Mass Transfer)

# Application: Upscaling of reactive transport processes in heterogeneous media



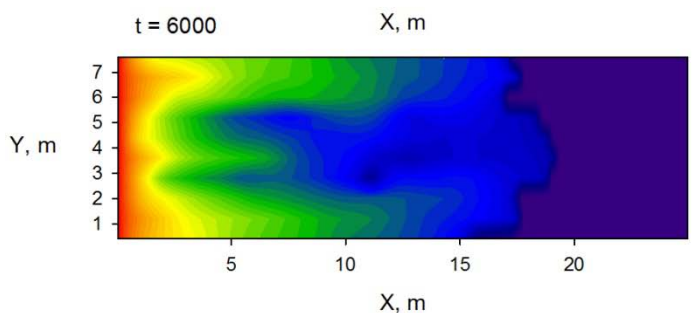
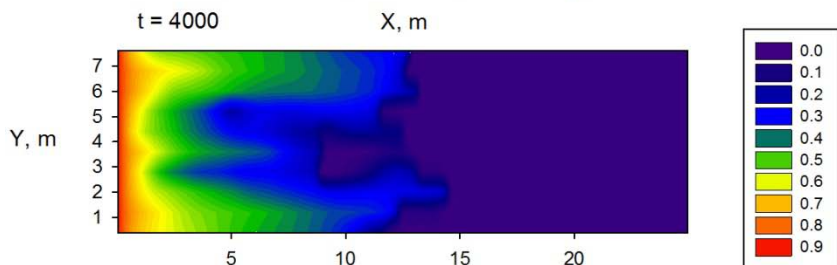
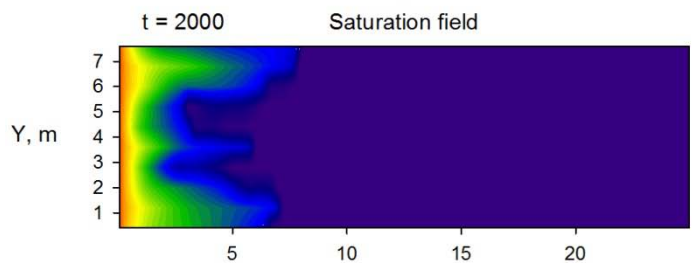
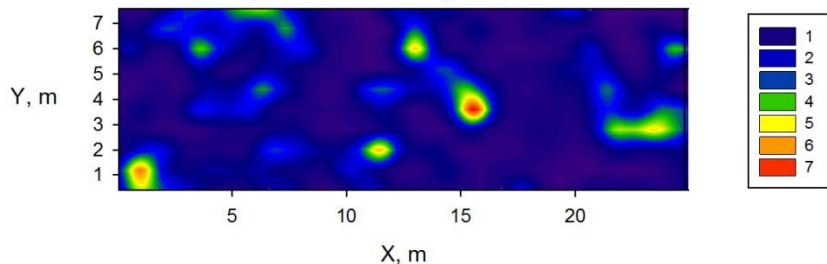
**(B)**



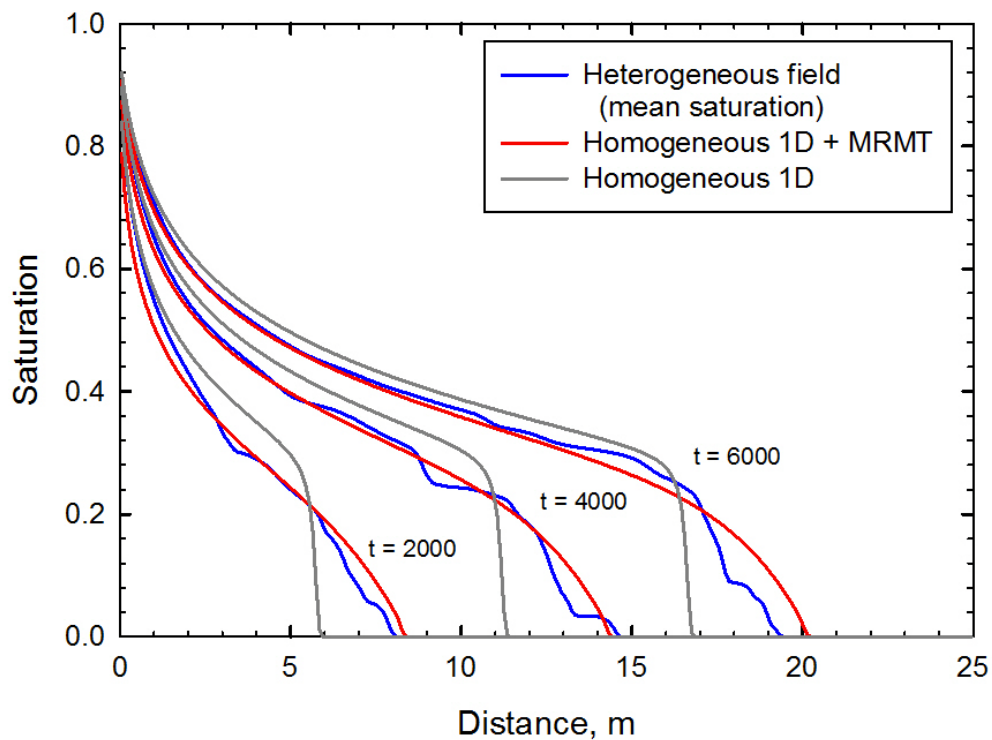
Scale effects in a system with a binary reaction of dissolution-precipitation equilibrium.: (A) Distribution of reaction rates and concentration of conservative component. (B) Comparison between a 2D heterogeneous model and a 1D effective model of MRMT-R.

# Application: Upscaling of multi-phase flow in heterogeneous media

Hydraulic conductivity field,  $\times 10^{11}$

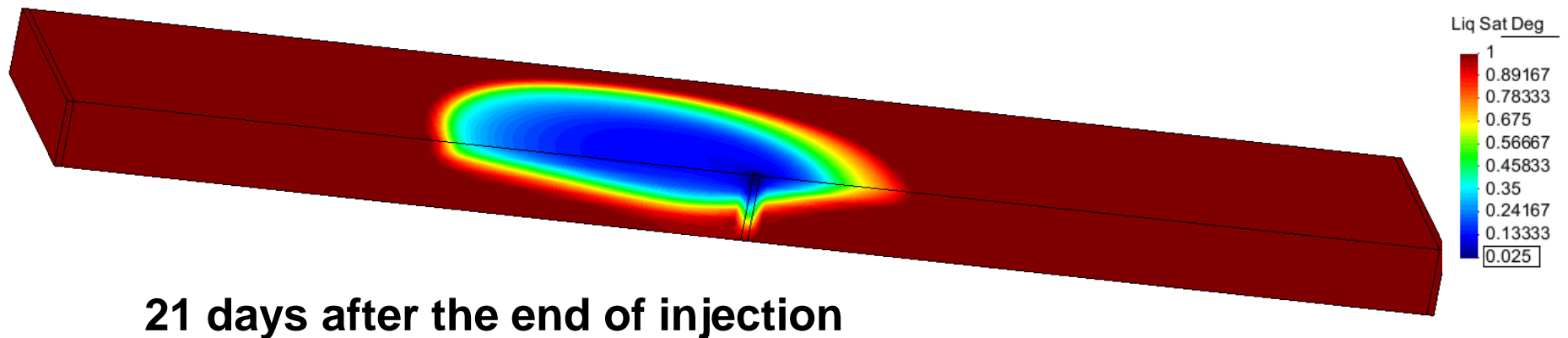
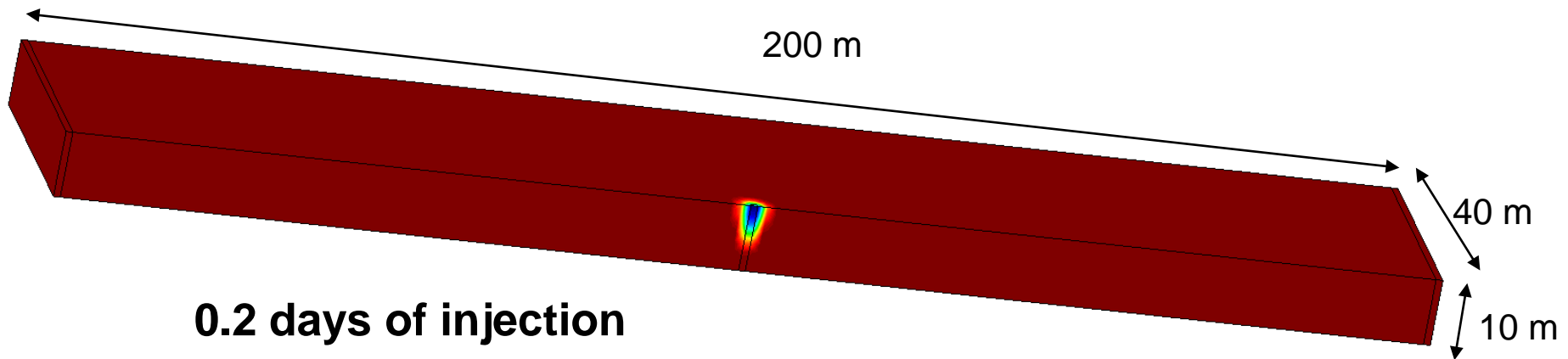


## Buckley-Leverett problem 2D - Heterogeneous

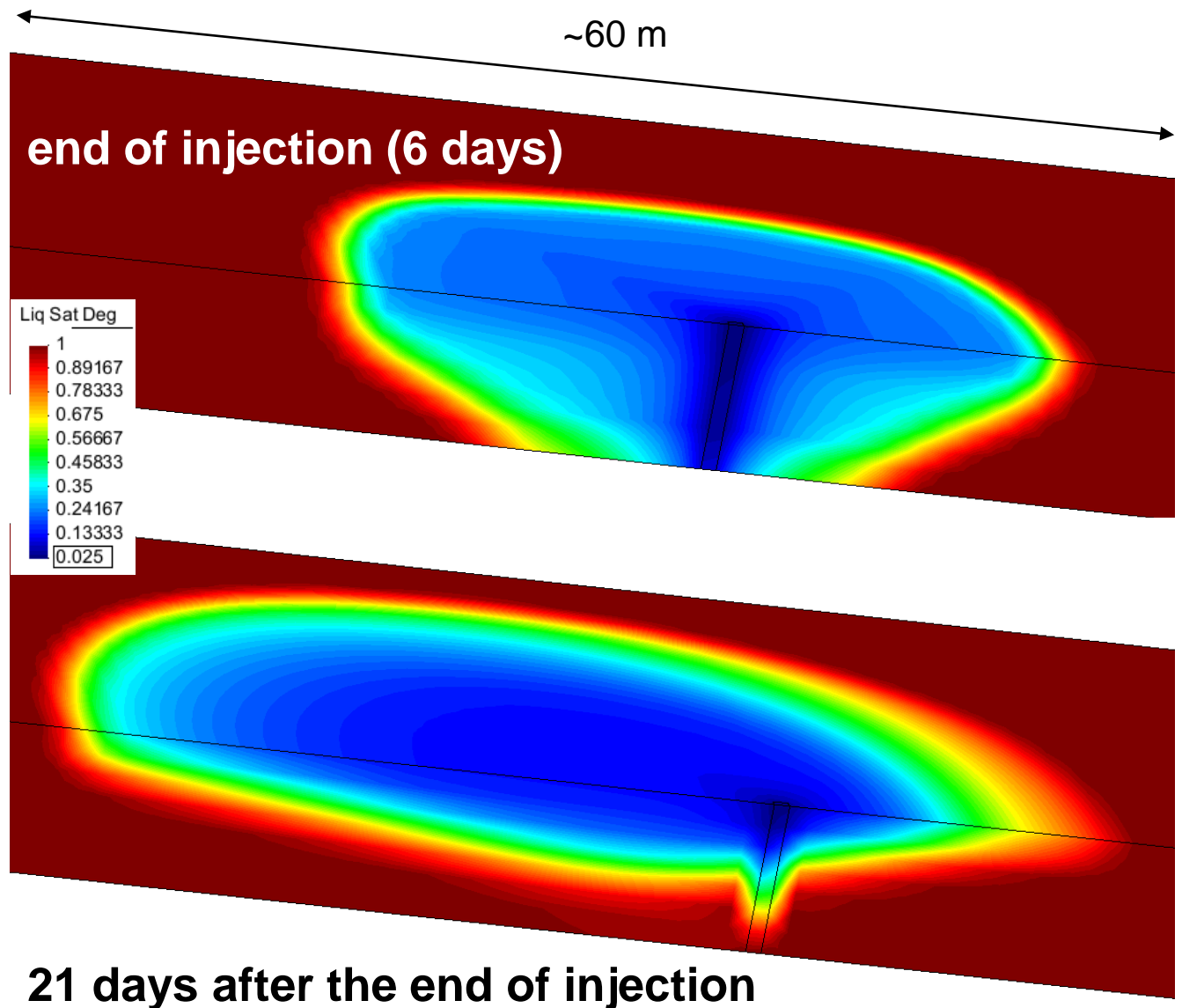


# CO<sub>2</sub> injection into anticline aquifers

CODE\_BRIGHT: 3D simulation of CO<sub>2</sub> injection at 1 kg/s into an aquifer with a slope of 5° (injection during 6 days)



# CO<sub>2</sub> injection into anticline aquifers

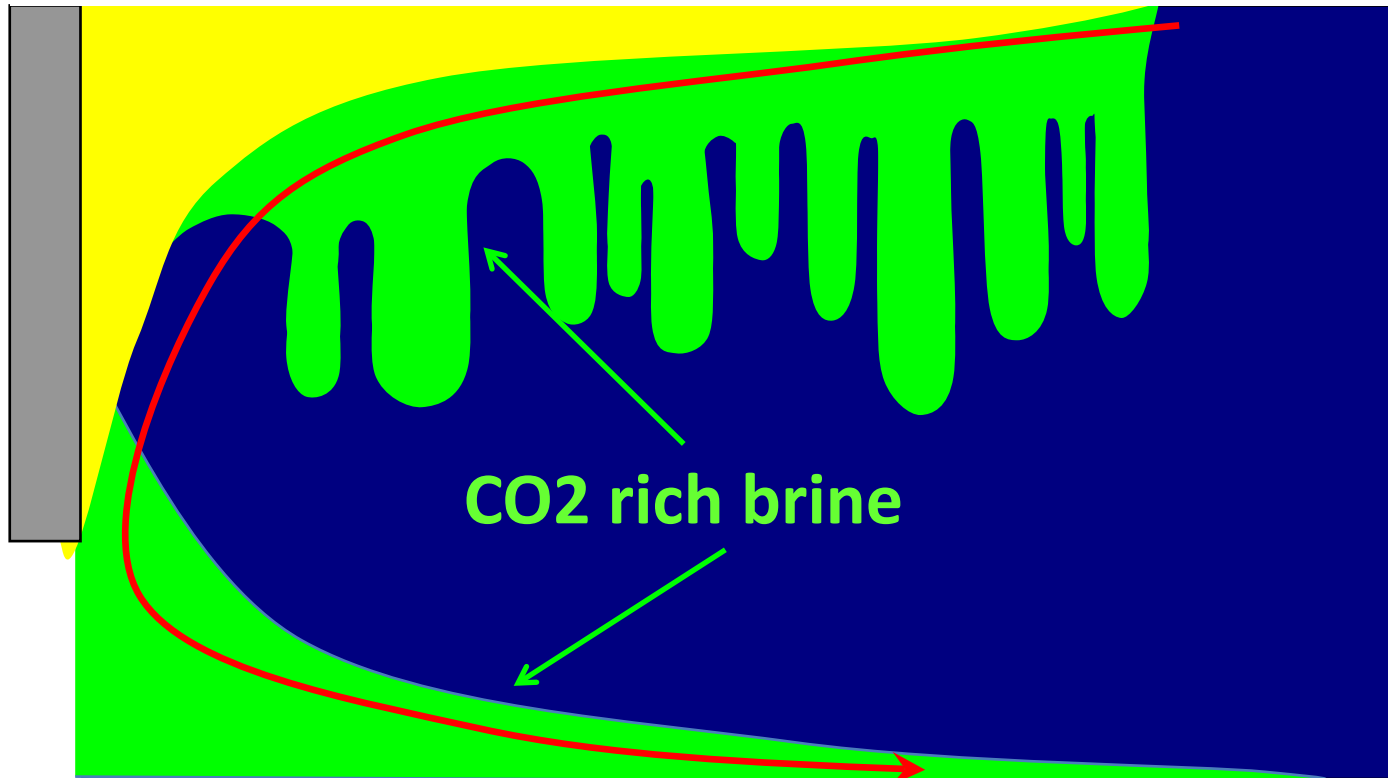


CO<sub>2</sub> buoyancy effect causes a migration of the CO<sub>2</sub> bubble upwards, specially after injection stop.

Gravitational forces dominate over viscous forces.

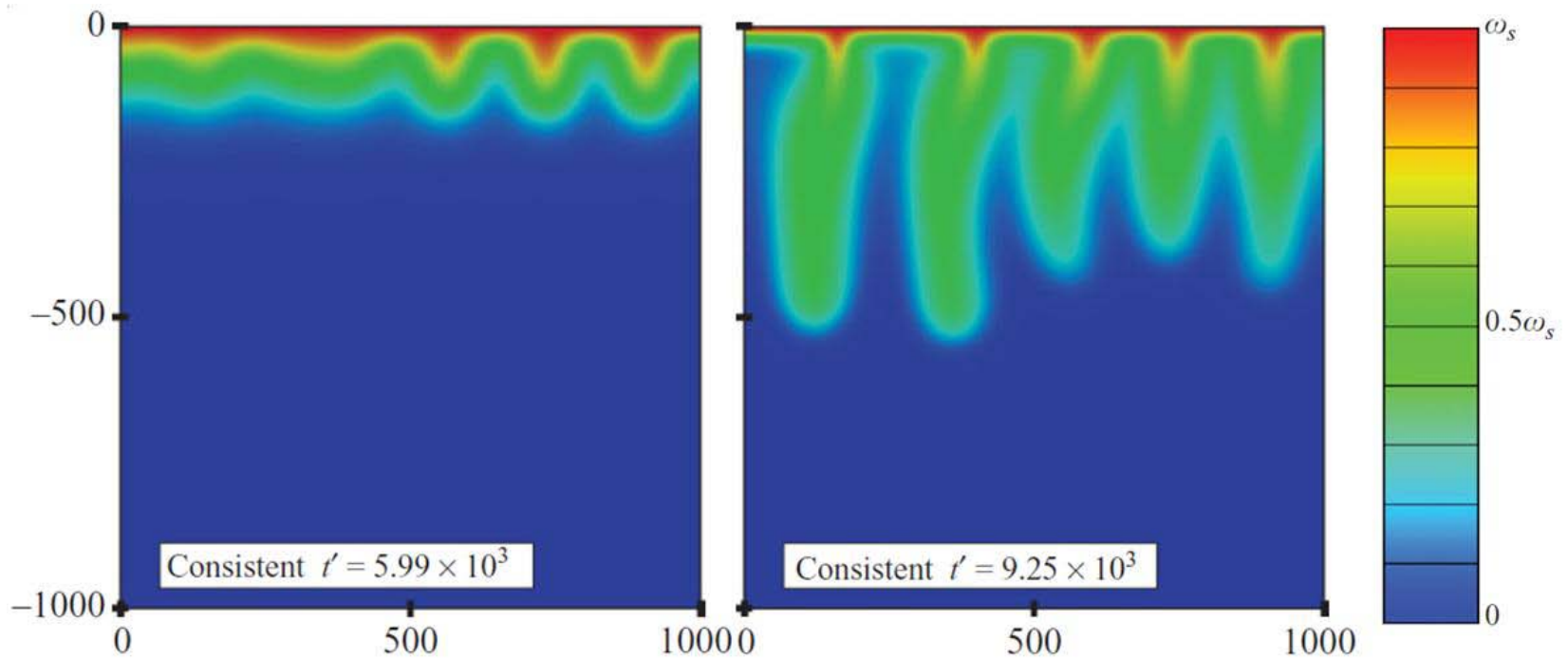
# CO<sub>2</sub> rich brine fingering

(CO<sub>2</sub> rich brine is denser and sinks in native brine)





# Scoping calculations on dissolution rate



# Scoping calculations on dissolution rate and onset of advection

$$m_s \approx 35u_b \Delta\rho \omega_s$$

$$u_b = k \Delta\rho g / \mu$$

$$t' = \frac{u_b}{\phi L_s} t$$

$$b_L = \frac{\alpha_L}{L_s}$$

$$L_s = \frac{\theta D_m + \alpha_L u_b}{u_b}$$

$\mu$	5,00E-04 Pa·s
$\Delta\rho$	25 kg/m <sup>3</sup>
k	1,00E-12 m <sup>2</sup>
ub	4,90E-07 m/s
$\omega_s$	2,50E-02
ms	1,07E-05 kg/m <sup>2</sup> /s
	<b>0.34 Mt/km<sup>2</sup>/yr</b>
$\alpha$	10 m
Ls	10
$\phi$	0,1
<b>t<sub>onset</sub></b>	<b>6,58 yrs</b>

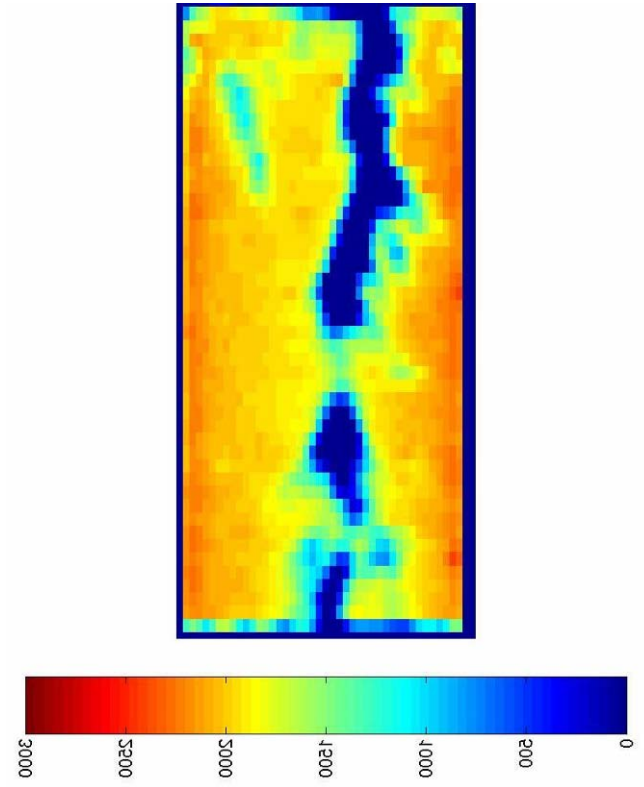
**Dissolution can be very significant after a few years** (which prevents further pressure buildup)

**Challenge: reduce the time for the onset of advection** (increase dispersion: viscous and capillary fingering, fluctuating injection)

# Chemical interactions

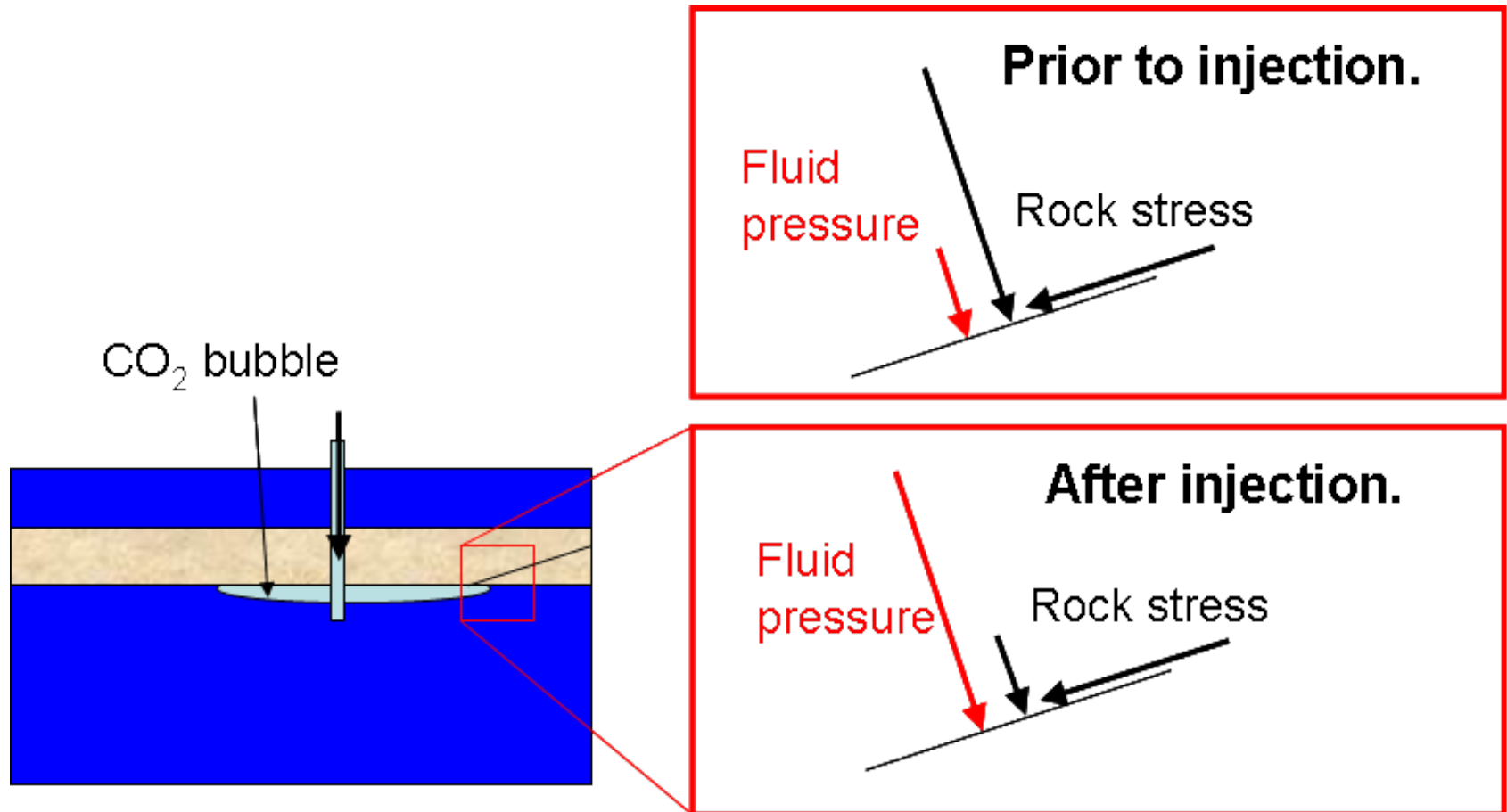
- CO<sub>2</sub> saturated brine will be very aggressive (pH around 3)
- This will cause:
  - Mineral dissolution
  - Partial neutralization of acidity (further solubility of CO<sub>2</sub>)
  - Depending on mineralogy of the rock and chemistry of the brine, other minerals may precipitate (e.g., if brine is rich in sulphate and rock rich in calcite, anhydrite will precipitate)
- Mineral dissolution increases the permeability of the rock

# Chemical coupling



Core photos prior to and (left) and after (center) injecting a CO<sub>2</sub> rich solution and tomographic image (right). (Wormholes)

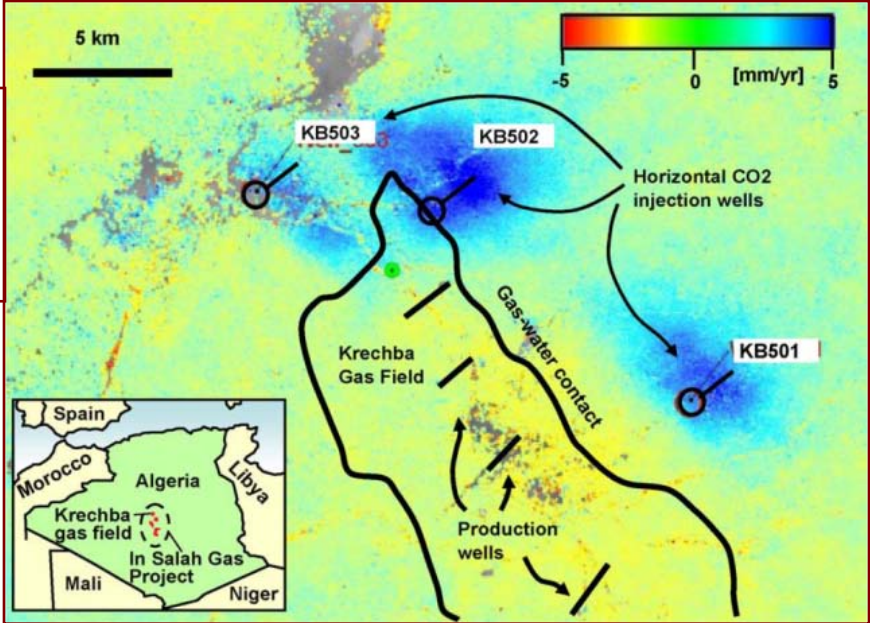
# Mechanical coupling\_ seal stability



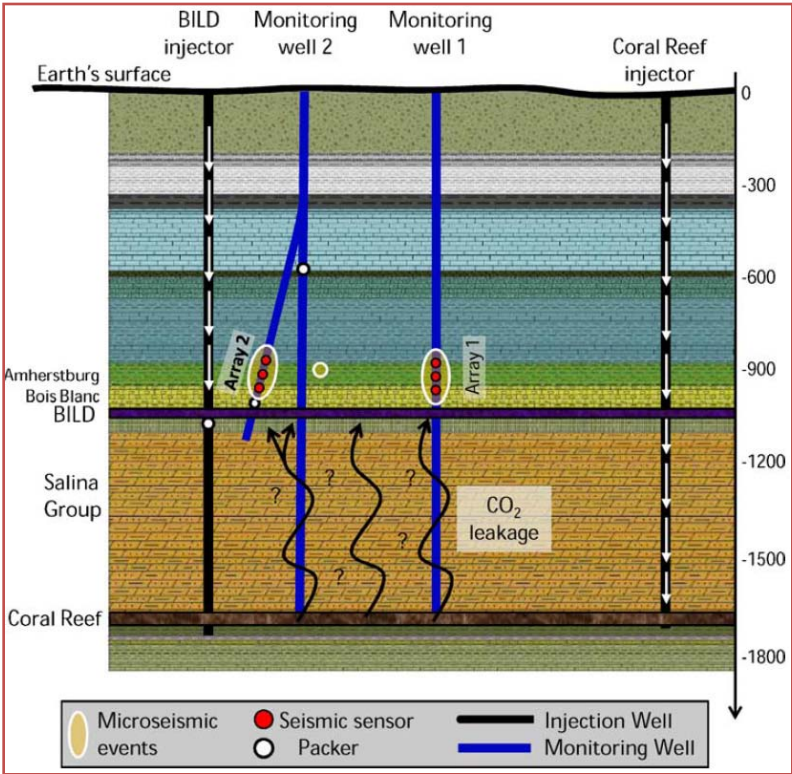
Pressure build up in the CO<sub>2</sub> bubble and surrounding brine may compromise mechanical stability of the cap rock and its isolating properties.

# HYDROMECHANICAL EFFECTS

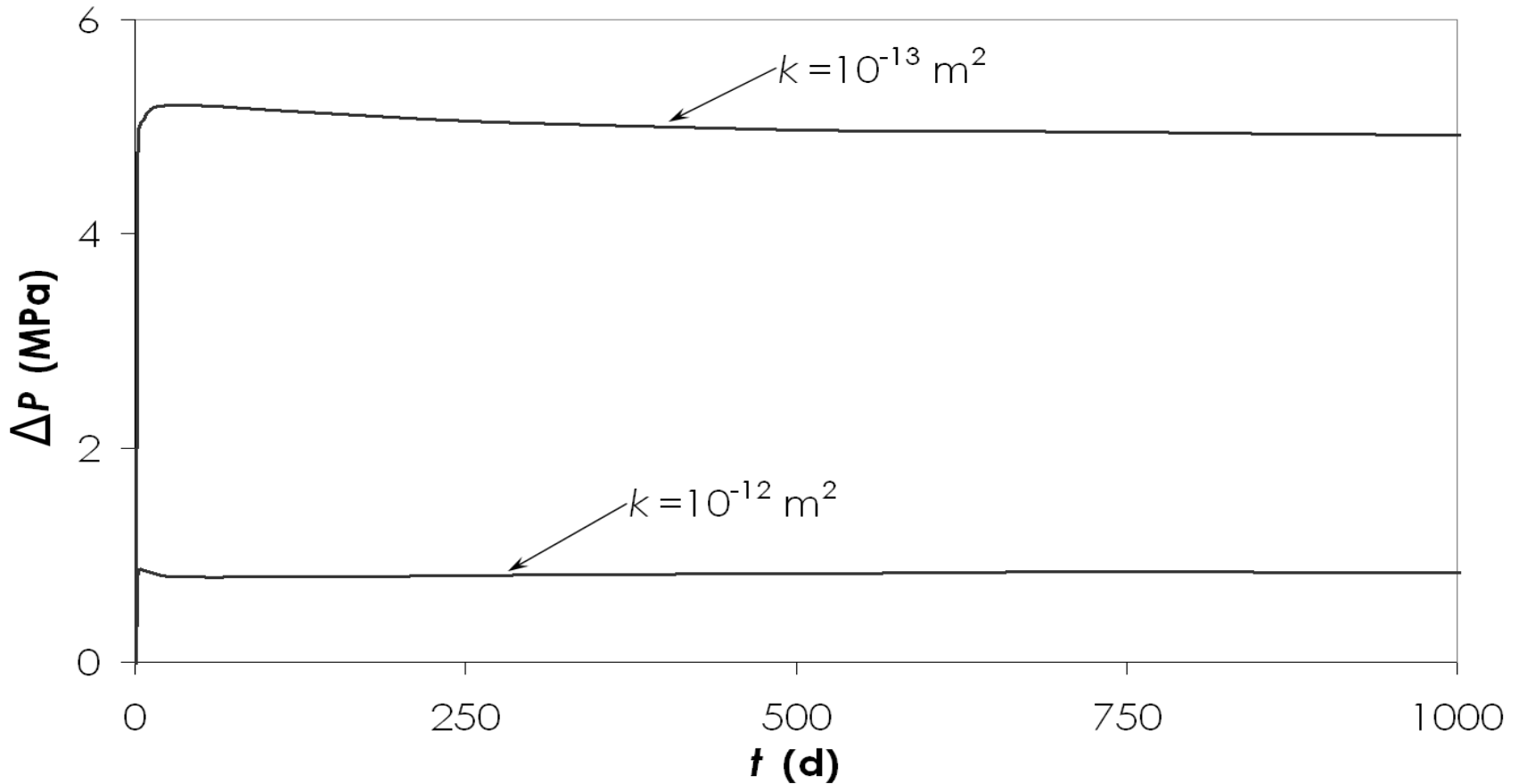
Uplift of 5 mm/yr in In Salah  
(Rutqvist *et al.*, 2010, IJGGC)



Microseismicity due to CO<sub>2</sub> leakage around wells in Otsego County, Michigan Basin, USA  
(Bohnhoff *et al.*, 2010, IJGGC)

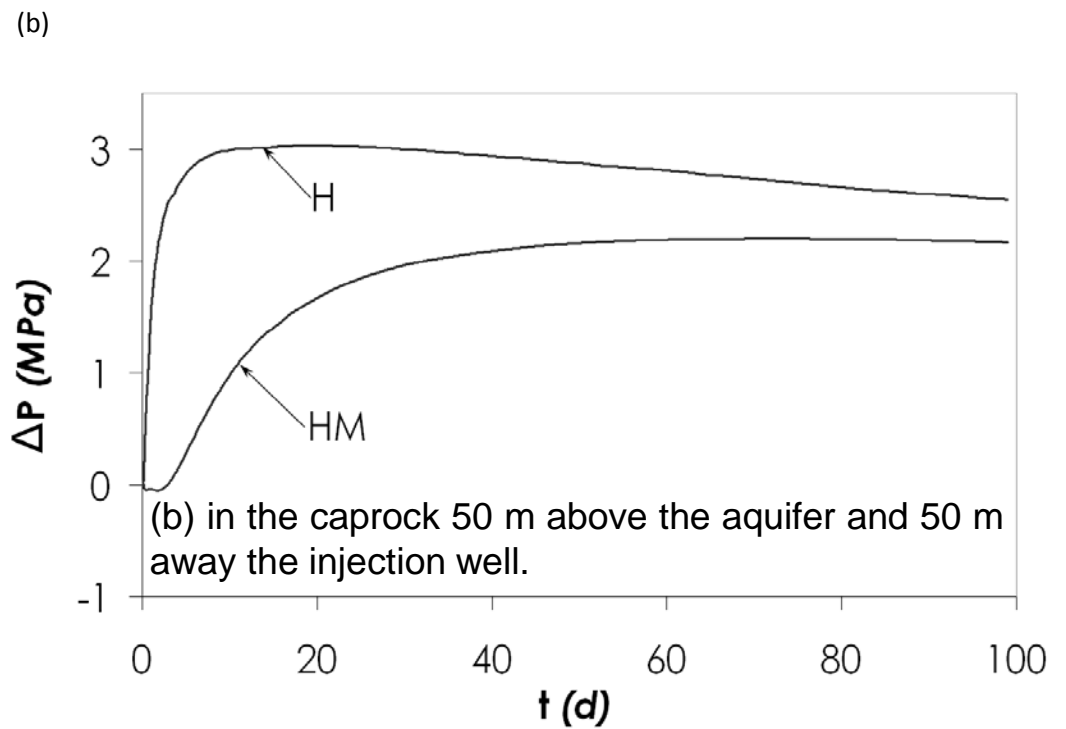
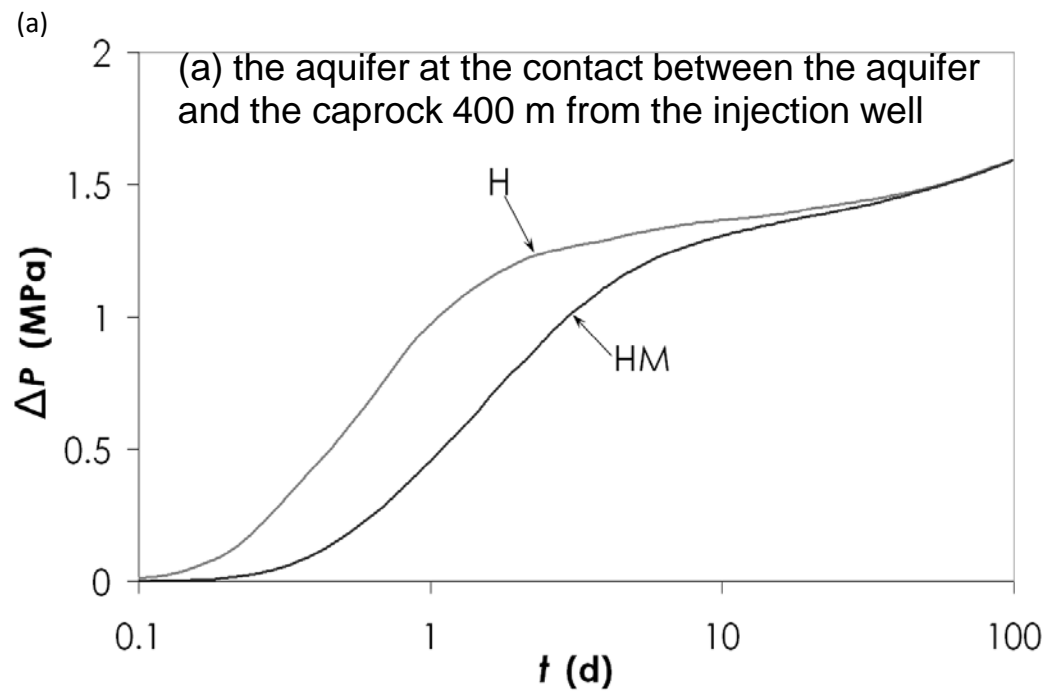


# Pressure buildup

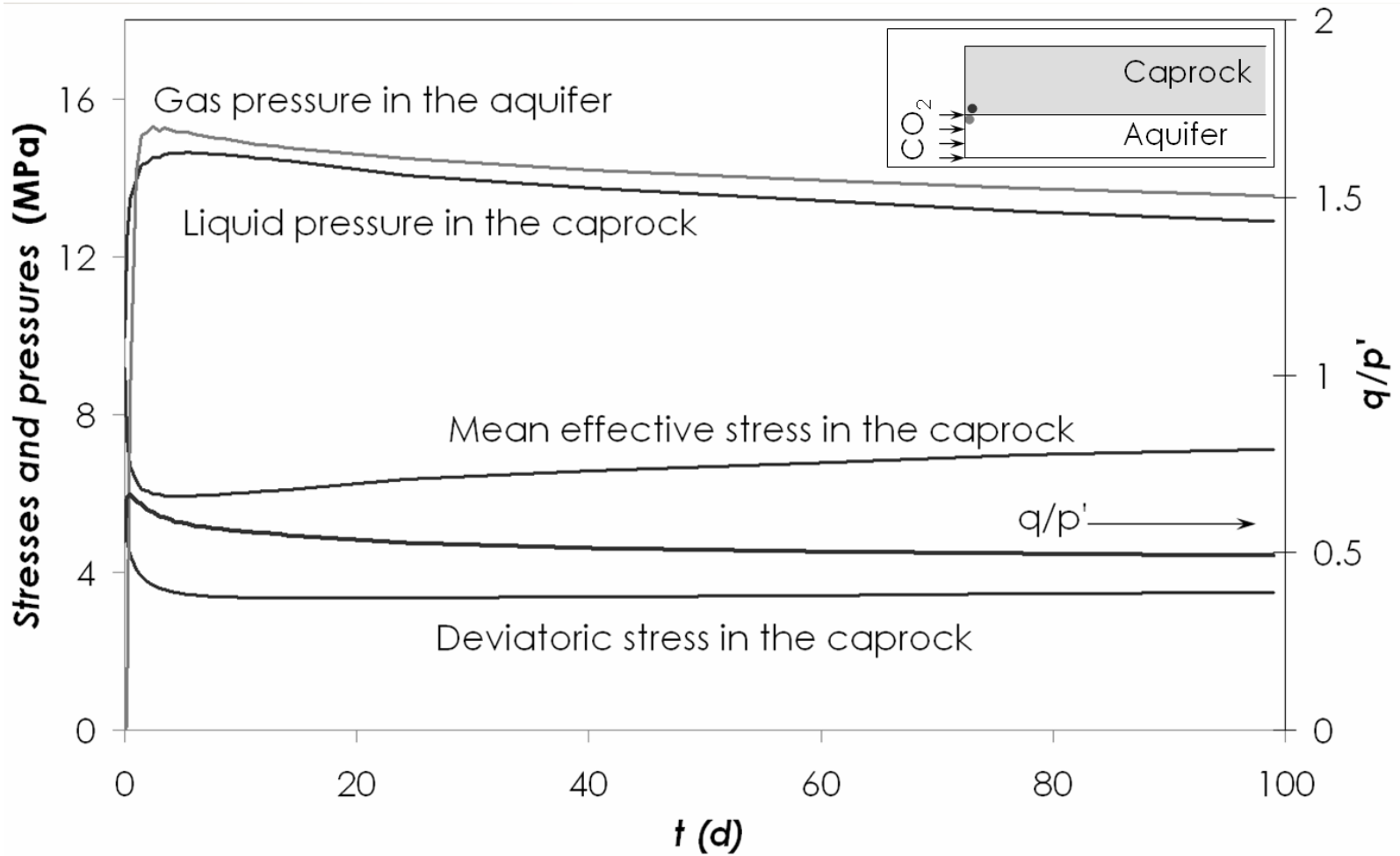


: Injection pressure at the top of the aquifer for a 1000 day injection period, for two intrinsic permeabilities of the aquifer. Injection pressure drops because of the lower  $\text{CO}_2$  viscosity with respect to that of brine.

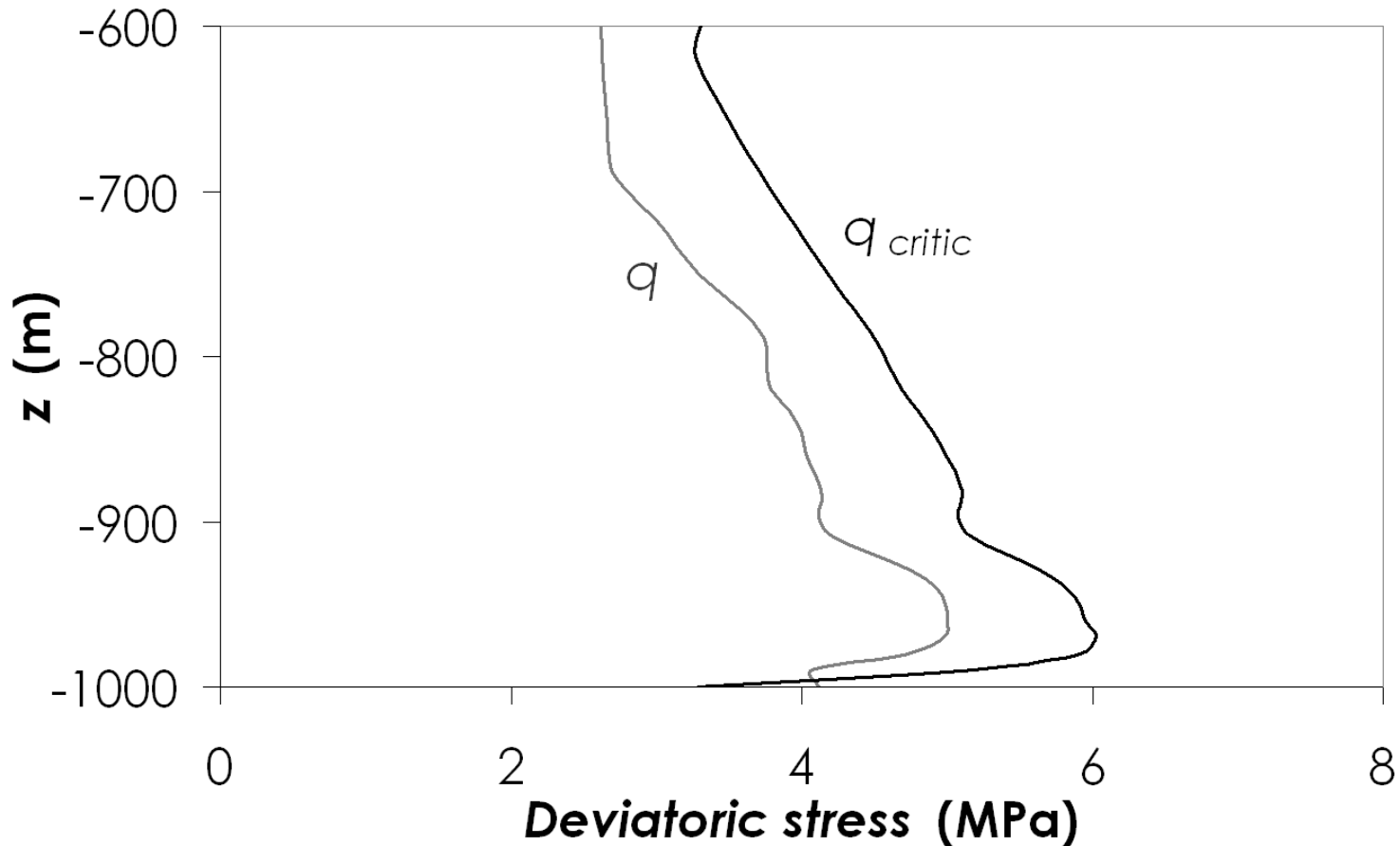
Fluid overpressure for a 100 days injection period, comparing pure hydraulic (H) with coupled hydromechanical (HM) simulation in and



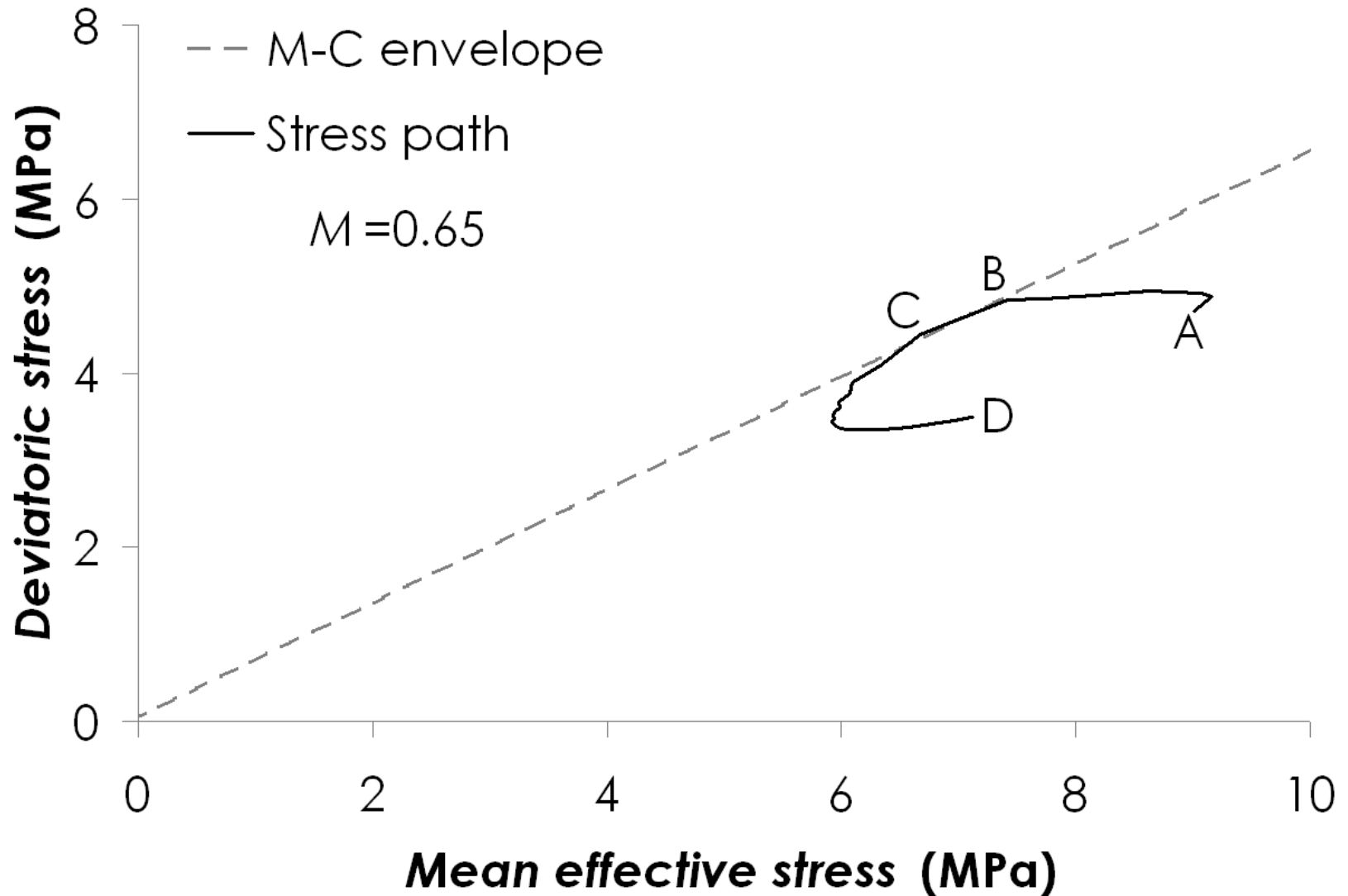




Stress and pressure evolution with time at the beginning of  $\text{CO}_2$  injection at the base of the caprock next to the injection well (see location in inset).



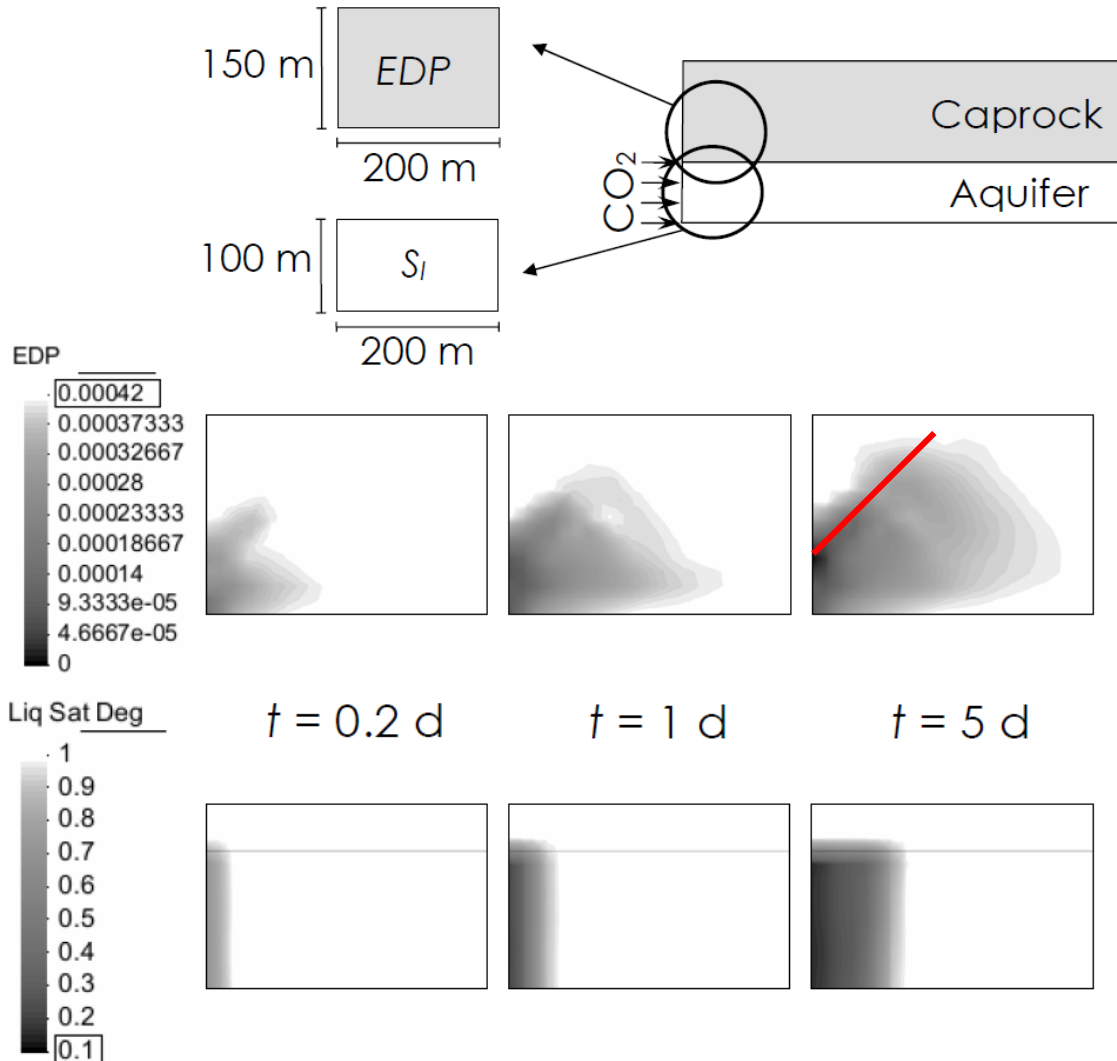
A vertical section of the caprock in the vicinity of the well, with the deviatoric stress,  $q$ , and critical the deviatoric stress,  $q_{critic}$ , after 4 days of injection. Note that  $q$  exceeds  $q_{critic}$  in the contact between the caprock and the aquifer, thus causing the caprock to plastify.



:  $(q, p')$  trajectory for a 100 day injection. The initial and final states are represented by A and D, respectively. The onset of plasticity takes place during early times (B-C), but plastic behavior eventually stops.

# Hydro-mechanical coupling

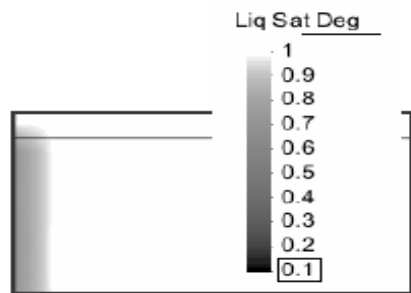
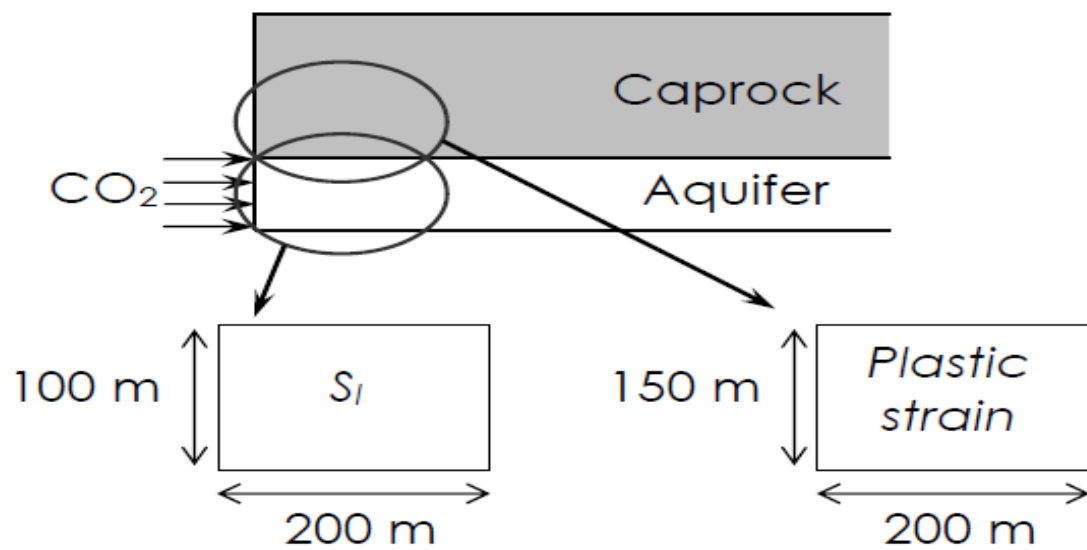
## Evolution of deformations and CO<sub>2</sub> saturation



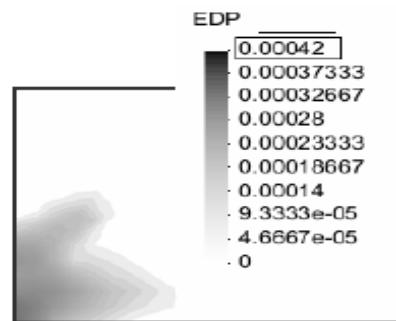
Caprock plasticize as the CO<sub>2</sub> migrates (for a caprock relatively weak)

Plastic region propagates in a direction of 45°

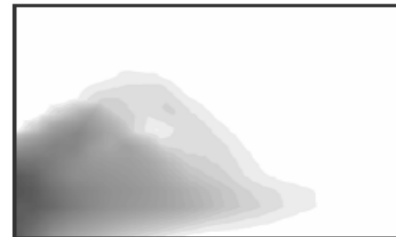
Plastic deformations stabilize after 5 days of injection



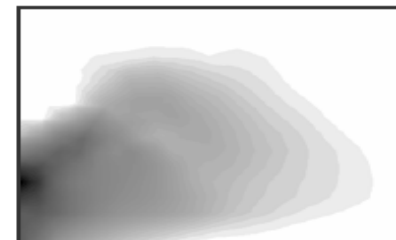
$t = 0.2 \text{ d}$



$t = 1 \text{ d}$



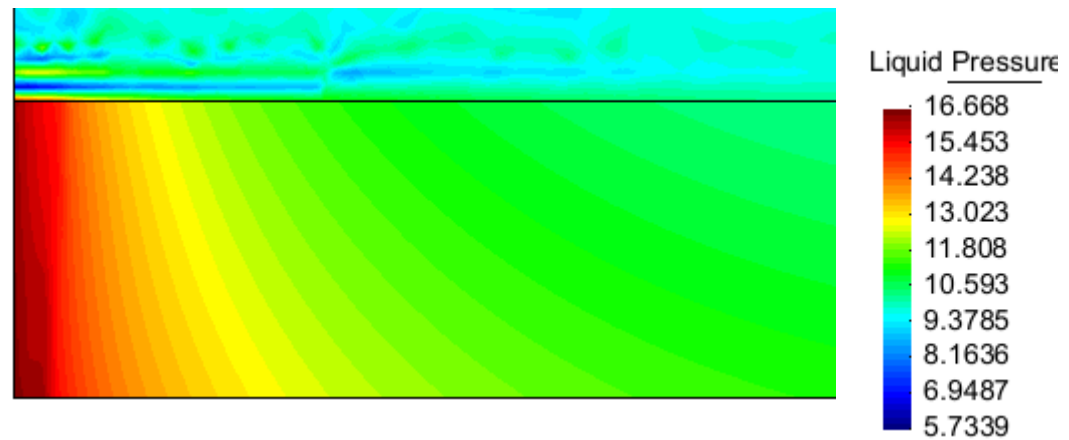
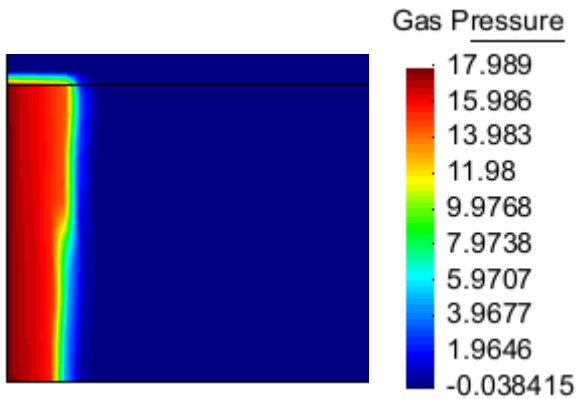
$t = 5 \text{ d}$



Plastic strain (*EDP*) in the caprock (left) and liquid saturation degree,  $S_l$ , in the aquifer (right) for different injection times. Plastic strain propagates as CO<sub>2</sub> advances at the beginning of injection.

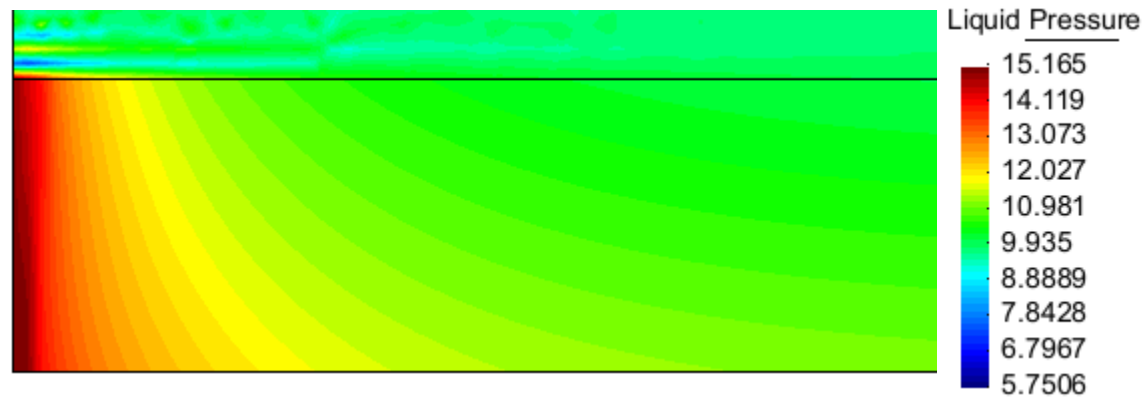
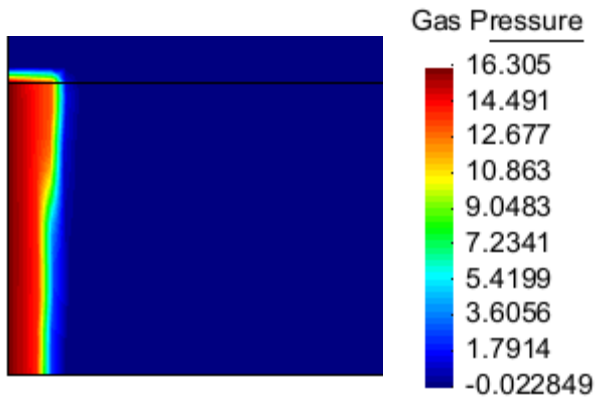
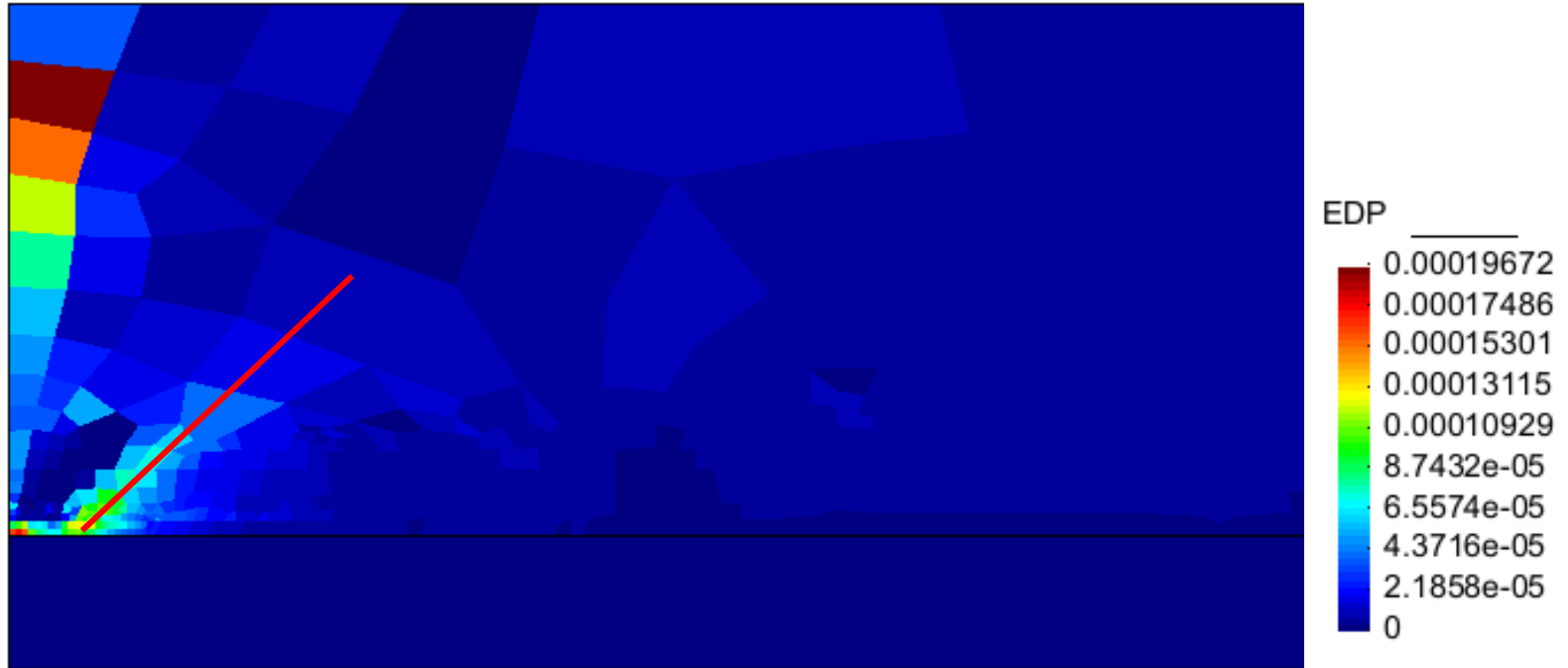
M=0.6

t=1.10703 d



M=0.4

t=1.09063 d



# Summary hydromechanical coupling

- Pressure buildup bounded
- HM coupling needed
- Shear conditions worst at the beginning
- Partially monitorable by measuring pressures at cap rock.
- Work on-going



# In summary

- Low density and viscosity
- High solubility
- Geochemistry
- Potential mechanical coupling
- Capillary effects
- Thermal effects
- Buoyancy
- Viscous fingering
- Gravity fingering
- Chemical fingering
- Seal Rock failure risk
- Migration of CO<sub>2</sub> and/or brine
- and, CCS expensive

TheHontomin challenge:

¿Can we understand all this qualitatively?

¿Can we take advantage of these “problems” to reduce costs?

# Key issues

## **Reduce costs**

Reduce Energy:

Permeability how much?, how to increase it?

Increase storage capacity

How much, how to increase it?

## **Reduce requirements**

Purity of CO<sub>2</sub>?

Increase safety

Improve social acceptance

## **Tool: Increase understanding**

# Key parameters

## Permeability: Controls

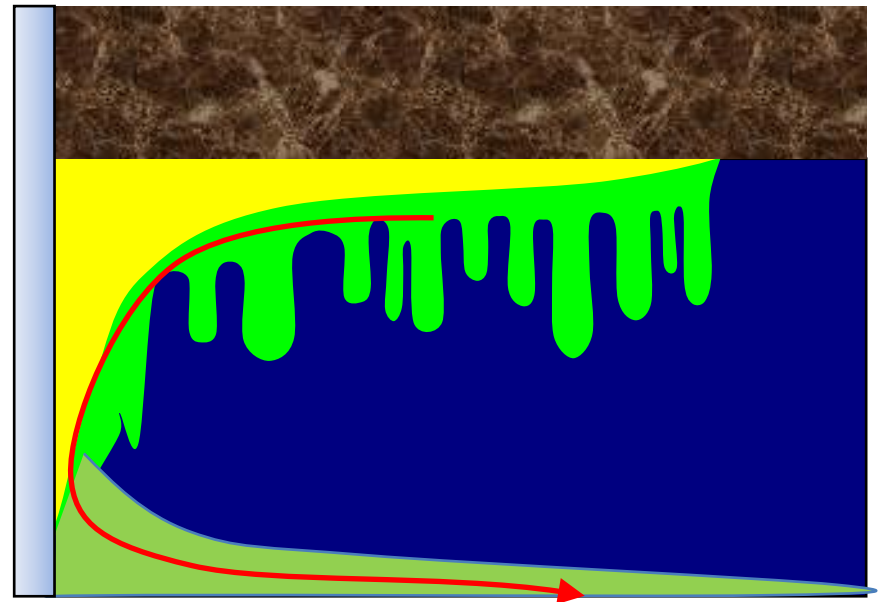
Effective: Buildup of injection pressure

Vertical: Rate of fingering (dissolution)

Connection: Migration of brine

Seal: vertical migration through seal

(But also entry pressure)



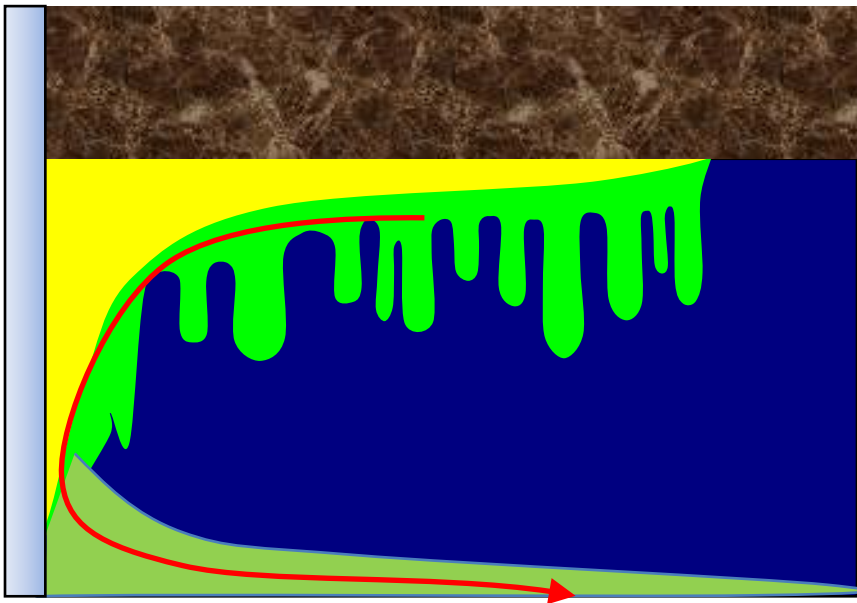
# Key parameters

## **Porosity:** Controls

Storage capacity

Porosity increase may increase permeability

But affect mechanical properties

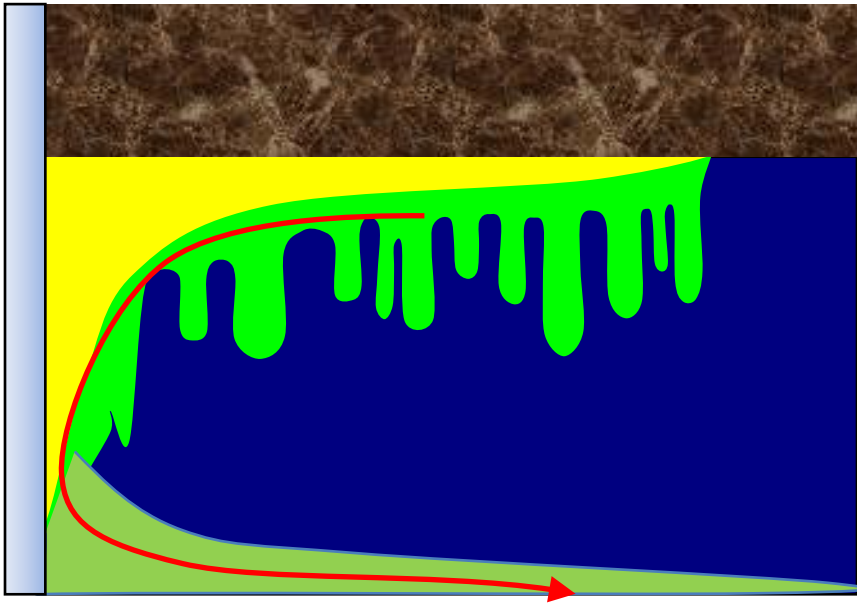


# Key parameters

## Reactivity: Controls

Porosity and permeability variations

State of CO<sub>2</sub>



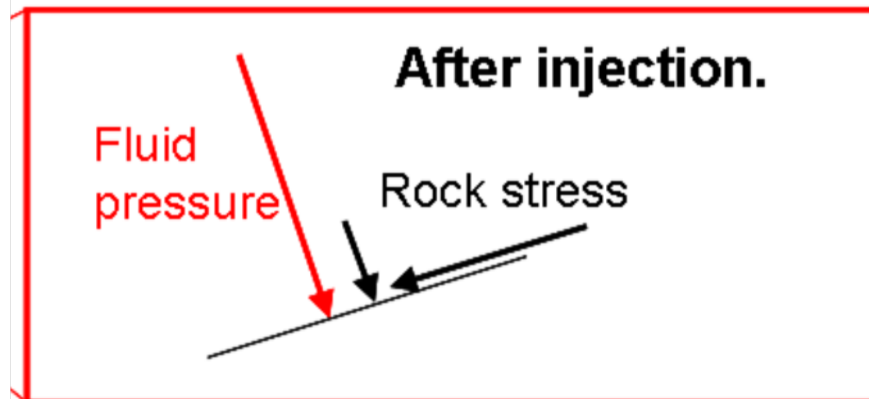
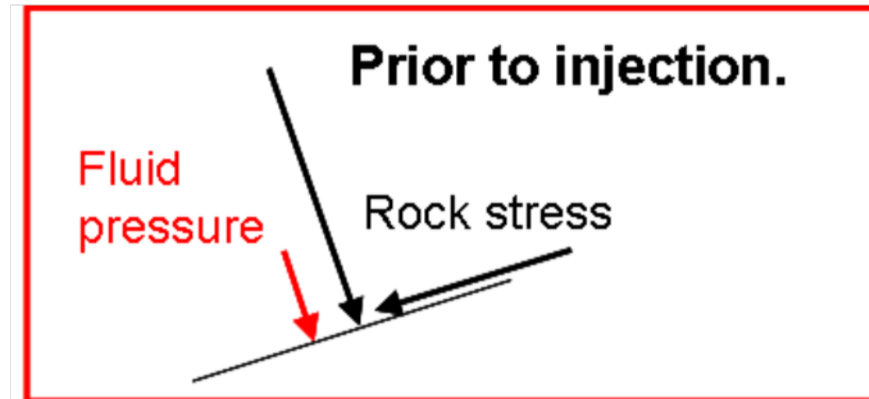
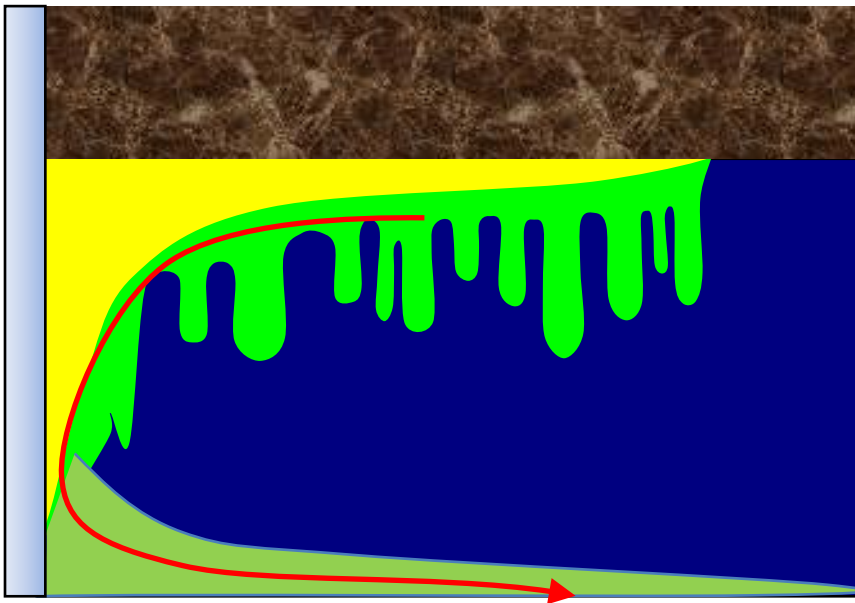
# Key issues

## Mechanical properties: Control

Compressibility: storage capacity

Long term safety

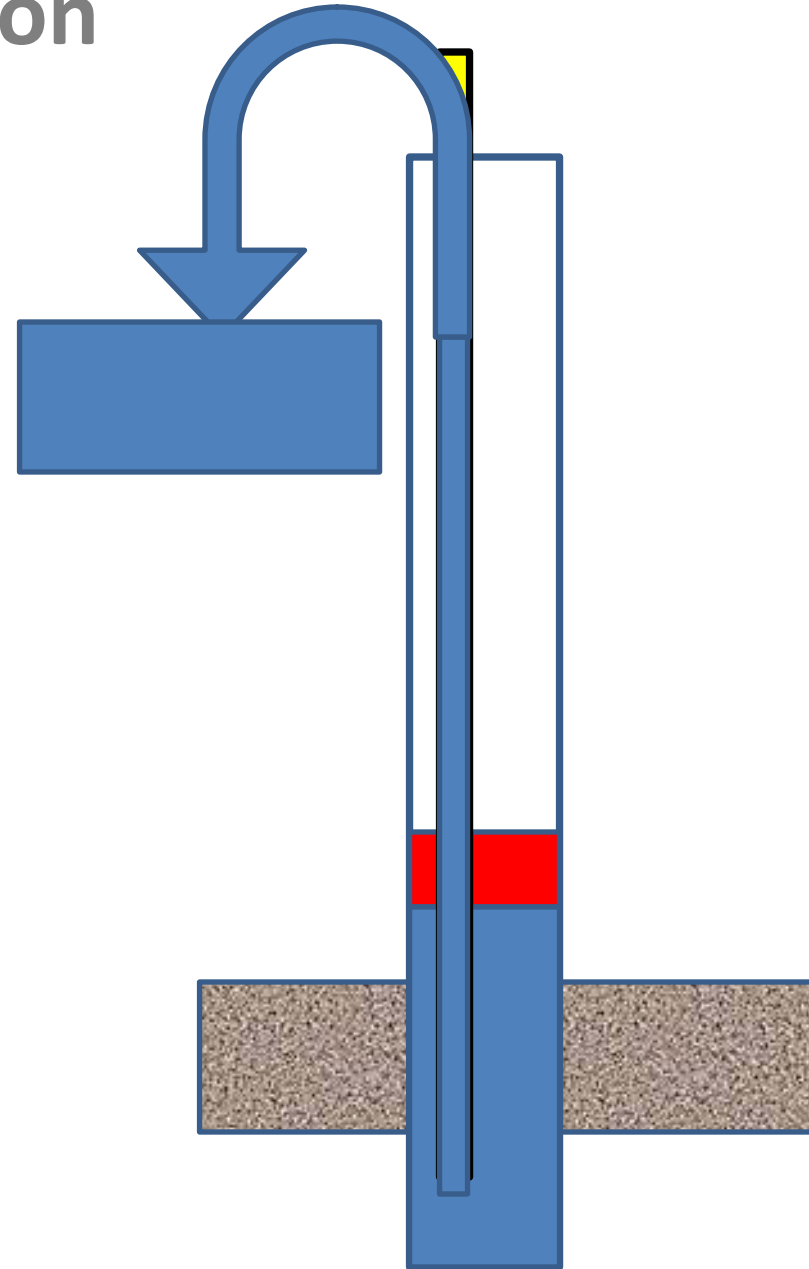
Social perception



# Water Injection-extraction

## 1. Pumping (Quita)

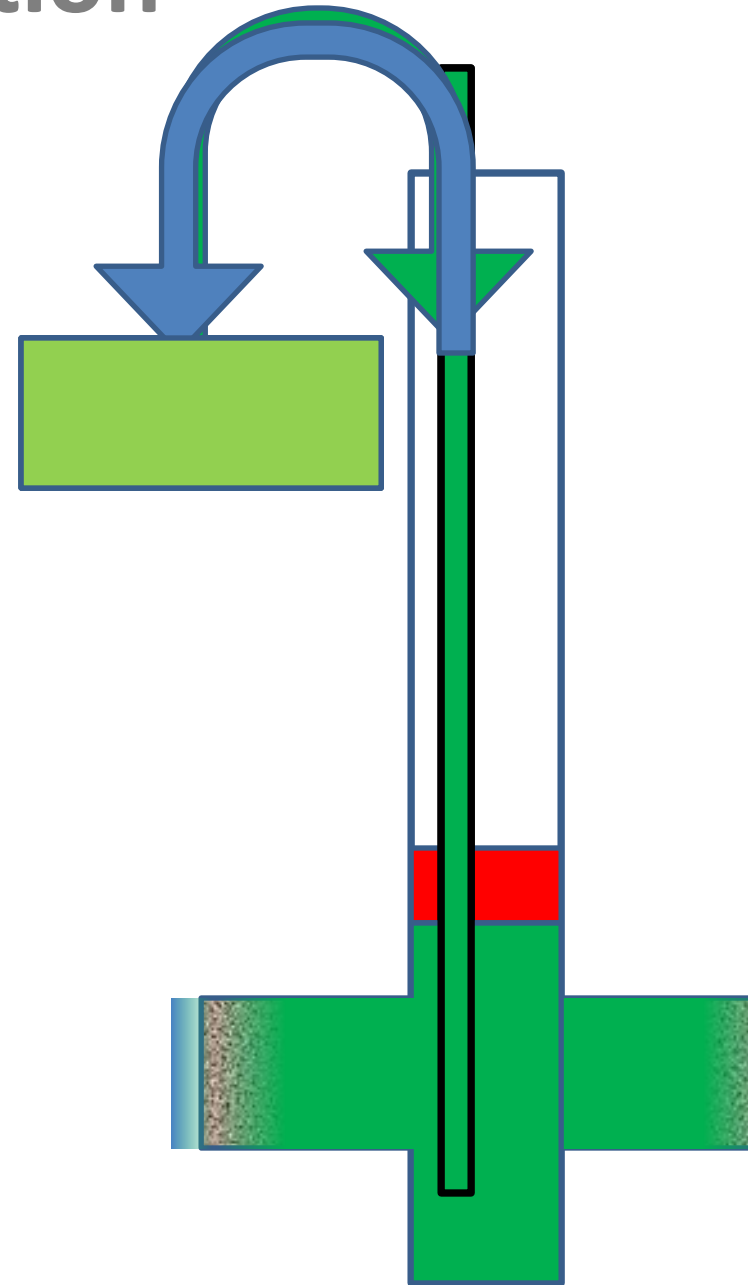
1. Pumping tests = 0.1-10L/s
2. Pumped water needs to be stored (conditioning and storage of up to 10000 m<sup>3</sup>)
3. Monitor drawdowns at all intervals to obtain  $K_{local}$ ,  $K_{eff}$ ,  $K_{vert}$ , and  $S$ . Also monitor deformation (but expect little).



# Water Injection-extraction

## 2. Injection (“y pon”)

1. Add tracer (and biocide)
2. Inject traced water
3. Rest
4. Extract water and monitor tracer(s) breakthrough. To obtain porosity structure, reactivity.
5. Repeat varying nature of tracers (conservative and reactive) and T
6. Repeat varying injection volume and rest time.





# CO2 push-pull (Mete-saca) test

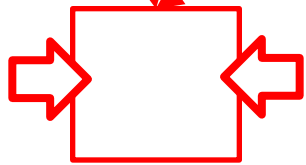
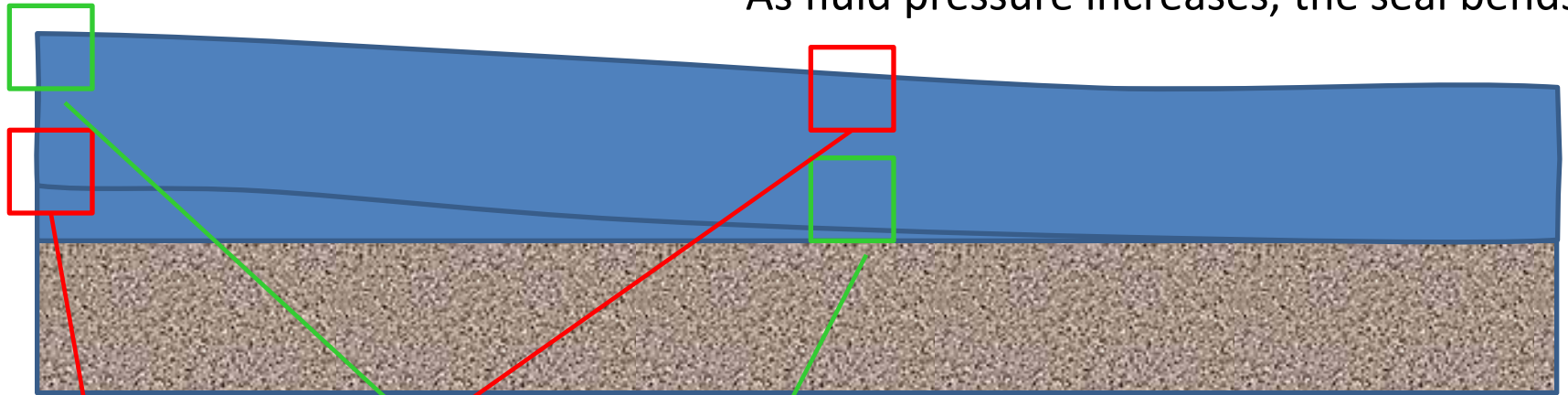
1. Inject CO2 (and gaseous tracers). Some 100 t de CO2
2. Electrical and thermal (heating) tests
3. Extract gas and evaluate mass of CO2 y concentration of gaseous tracers
4. Electrical and thermal tests
5. Informs about trapping mechanisms (specifically contact area and capillary trapped CO2)

# High pressure (and flow rate) injection test ("apretón")

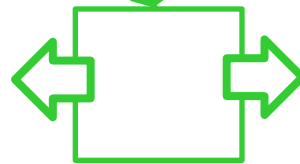
- Inject a high flow rate at a very high pressure (some 100 bar at surface)
- Observe pore fluid pressure, rock deformation and possible microseisms

# Hydro-mechanical coupling during “apretón”

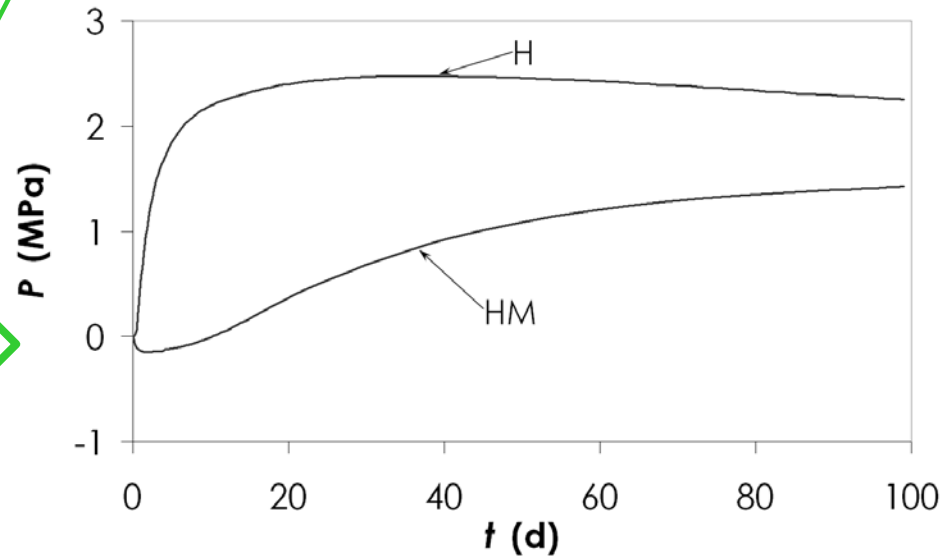
As fluid pressure increases, the seal bends



P goes up immediately



At first, P drops



# Heating test (“calentón”)

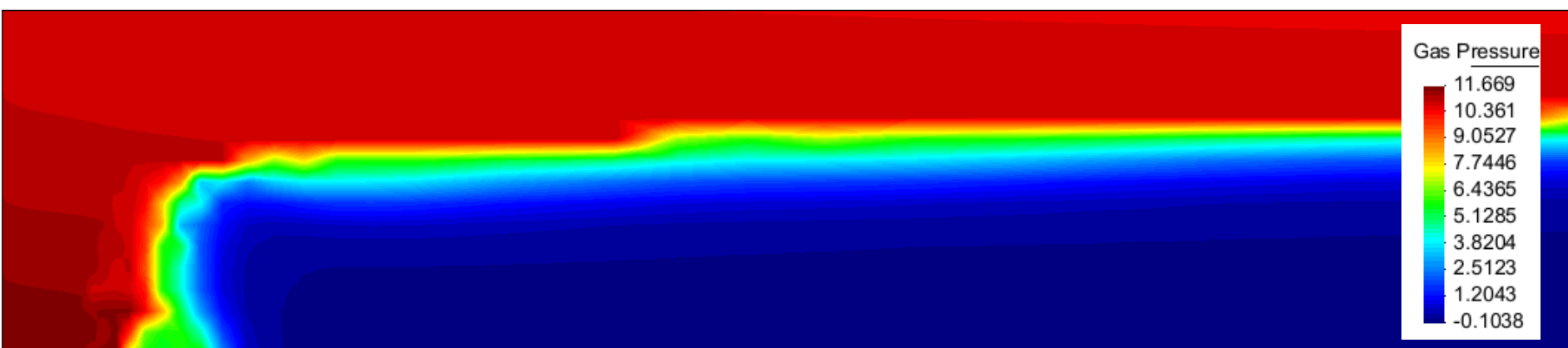
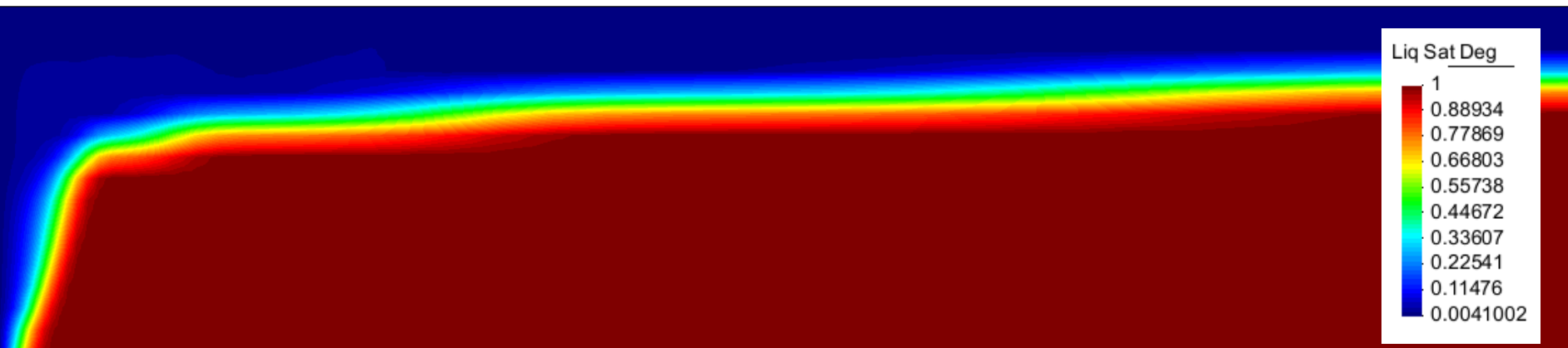
- Heat up the heater along the whole borehole
- Observe temperature buildup
- Deduce thermal capacity and conductivity
- Deduce presence of CO<sub>2</sub> (either continuous phase during injection tests, or capillary trapped during CO<sub>2</sub> “metasaca”)

# CO2 injection tests

- Conventional
- Fluctuating flow rate
- Liquid CO2 injection
- Others

# Disolución en la base del penacho da lugar a celdas de convección en la salmuera

La base del penacho será una fiesta (disolución de CO<sub>2</sub>, convección de salmuera, dis-precip mineral...)



**Thanks for your attention**